

Chemical Bonding and Molecular Structure

Name Deja Dinkins
 Instructor Mr. Berry Section 121L Partner Bau

Exercise I. Determining VSEPR Geometry. (Use balloons to complete the Table)
 (1/2 pt. for each box)

VSEPR Type	Lewis Structure	Number of Atoms Bonded to Atom A	Number of Lone Pairs Around Atom A	Geometry of Electron Clouds Around A	Molecular Geometry of Atoms Around A (Molecular Shape)	Bond Angle B-A-B
AB_2	$B-A-B$					
AB_3	$\begin{array}{c} B-A-B \\ \\ B \end{array}$					
AB_2E	$\begin{array}{c} \cdot \\ B-A-B \end{array}$					
AB_4	$\begin{array}{c} B \\ \\ B-A-B \\ \\ B \end{array}$					
AB_3E	$\begin{array}{c} \cdot \\ B-A-B \\ \\ B \end{array}$					
AB_2E_2	$\begin{array}{c} \cdot \\ B-A-B \\ \\ \cdot \end{array}$					

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