

ChemCom 1st Semester Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following observations would predict a physical change?
a. rusting of a nail
b. ignition of paper
c. emission of a gas
d. evaporation of water
- _____ 2. Which of these is a chemical property?
a. resistance to corrosion
b. malleability
c. boiling point
d. solubility in water
- _____ 3. Properties such as melting point, density, and electrical conductivity are examples of
a. chemical properties
b. nuclear properties
c. physical properties
d. electrical properties
- _____ 4. All of the following properties are physical properties of a substance except
a. boiling point
b. hardness
c. malleability
d. combustibility
- _____ 5. What two measurements of a substance make up the substance's density value?
a. mass and weight
b. volume and pressure
c. mass and volume
d. temperature and mass
- _____ 6. A physical property of a substance
a. can be determined without changing the identity of the substance
b. depends on the way a substance reacts
c. is a property that helps support life
d. involves breaking the chemical bonds that hold compounds together
- _____ 7. A change in which bonds are broken and new bonds are formed, resulting in new substances being produced is called
a. a phase change
b. a nuclear change
c. a chemical change
d. a physical change
- _____ 8. Which of these observations is evidence of a chemical change?
a. A drink mix dissolves in water.
b. Bubbles of a gas explode when ignited.
c. Rubbing alcohol evaporates from a surface.
d. A pop can is crushed by a hammer.
- _____ 9. All of the following are chemical changes except
a. Wood burns in a fireplace
b. A glass of milk becomes sour.
c. Mold "grows" on cheese.
d. Sugar dissolves in water.
- _____ 10. Which of these is a chemical change?
a. Tomato juice is made from ripe tomatoes.
b. A pencil is sharpened.
c. A brass doorknob tarnishes.
d. Friction wears away the tread on a tire after driving it many miles.
- _____ 11. The following statement contains five numbered properties of chlorine:
Chlorine is a yellow gas (1) that is more dense than air (2). It forms hydrochloric acid when breathed into the lungs (3) and kills bacteria (4) when dissolved in water (5).
How many of the five properties in the statement are physical properties?
a. 1
b. 2

Chemcom Guide

Rachel Sandford



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Chemistry in the Community American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design *Innovative Curriculum Materials* ,1999 *ENC Focus* ,1999 **H.R. 4271, the National Science Education Act; H.R. 4272, the National Science Education Enhancement Act; and H.R. 4273, the National Science Education Incentive Act** United States. Congress. House. Committee on Science, 2001

Handbook on the Physics and Chemistry of Rare Earths: without special title ,1986 **Science Curriculum Resource Handbook** ,1992 **Chemistry in the Community** ,1995 **El-Hi Textbooks and Serials in Print** ,2006 **Chemists' Guide to Effective Teaching** Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 Part of the Prentice Hall Series in Educational Innovation for Chemistry this unique book is a collection of information examples and references on learning theory teaching methods and pedagogical issues related to teaching chemistry to college students In the last several years there has been considerable activity and research in chemical education and the materials in this book integrate the latest developments in chemistry Each chapter is written by a chemist who has some expertise in the specific technique discussed has done some research on the technique and has applied the technique in a chemistry course **Analyzing the Curriculum** George J. Posner, 1995 As a primary text this book provides the backbone for a basic curriculum course at either graduate or undergraduate level It attempts to teach how the parts of a curriculum fit together and how to identify assumptions underlying curricula In this way students develop the ability to determine why a curriculum proves better for some than for others what approaches to teaching are compatible with a particular curriculum what difficulties a curriculum is likely to encounter during implementation and what kinds of changes in the curriculum parents students and administrators are likely to demand These are important skills for evaluating selecting and adapting existing programmes to suit particular situations *Focus a Magazine for Innovators* ,1999 *Science and Mathematics Education in the United States* Organisation for Economic Co-operation and Development, 1993 On cover title page OECD documents **Illinois Chemistry Teacher** ,2001 *El-Hi Textbooks & Serials in Print*, 2005 ,2005 *Chemunity News* ,1993 Newsletter for chemistry educators at the elementary high school and college levels **ChemCom** American Chemical Society, 1993 **Mathematics and Science Curriculum Programs** ,2002 **The Science Teacher** ,2001 The American Chemical Society at 125 ,2002 *Celebrating 125 Years of the American Chemical Society* ,2001

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