

General Chemistry placement test Study Guide Exam with 100% Verified Solutions

molecule - answer ✓ ✓ two nonmetal atoms combined form a covalent compound called a diatomic molecules - answer ✓ ✓ H₂, O₂, F₂

Matter - answer ✓ ✓ Anything that has mass and takes up space

Element - answer ✓ ✓ A substance that cannot be broken down into simpler substances

Compound - answer ✓ ✓ A pure substance made of two or more elements CHEMICALLY combined.

Compound - answer ✓ ✓ Can be decomposed by chemically

Mixture - answer ✓ ✓ A PHYSICAL blend of two or more components.

(aq) - answer ✓ ✓ aqueous mixture, a substance is dissolved in water

homogeneous mixture - answer ✓ ✓ A mixture that is so evenly mixed that you can't see all of the different parts. Ex. Air, kool-aid

heterogeneous mixture - answer ✓ ✓ A mixture in which different materials can be seen easily.

Ex. soil, chocolate chip cookies

filtration - answer ✓ ✓ process to separate a solid "precipitate" from a liquid

sublimation - answer ✓ ✓ phase change from (s) → (g)

empirical formula - answer ✓ ✓ A formula showing the lowest whole number ratio of atoms in a compound. Can't be reduced.

solid - answer ✓ ✓ A state of matter that has a definite shape and a definite volume

liquid - answer ✓ ✓ A form of matter that flows, has constant volume, and takes the shape of its container

Bromine (Br) and Mercury (Hg) - answer ✓ ✓ only two liquids on the periodic table

gas - answer ✓ ✓ A state of matter with no definite shape or volume

gas - answer ✓ ✓ will uniformly and completely fill a container

Chemistry Placement Exam Study Guide

Mary Shelley

Chemistry Placement Exam Study Guide:

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