

Answer Sheet for Sample Problems for Chemistry Level 1 Final Exam 2013-2014 Study Guide

Electrons in Atoms – Chapter 5

1. What is the frequency of green light, which has a wavelength of $4.90 \times 10^{-7} \text{ m}$?

$$c = \lambda \cdot v \quad \lambda = 4.90 \times 10^{-7} \text{ m}, c = 3.00 \times 10^8 \text{ m/s} \quad v = ? \text{ Hz} \quad v = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{4.90 \times 10^{-7} \text{ m}}$$

$6.12 \times 10^{14} \text{ Hz}$

2. An x-ray has a wavelength of $1.15 \times 10^{-10} \text{ m}$. What is its frequency?

$$c = \lambda \cdot v \quad \lambda = 1.15 \times 10^{-10} \text{ m}, c = 3.00 \times 10^8 \text{ m/s} \quad v = ? \text{ Hz} \quad v = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{1.15 \times 10^{-10} \text{ m}}$$

$2.61 \times 10^{28} \text{ Hz}$

3. Which type of electromagnetic radiation has the most energy: Microwaves, X-Rays, or Visible light? Why?

Ans: X-Rays, b/c they have the shortest wavelength and highest frequency of the types listed and frequency and energy are directly related.

4. What is the energy of a $9.50 \times 10^{13} \text{ Hz}$ radiation?

$$E = hv \quad v = 9.50 \times 10^{13} \text{ Hz}, h = 6.626 \times 10^{-34} \text{ Js} \quad E = ? \text{ J}$$

$$E = hv = (6.626 \times 10^{-34} \text{ Js}) \times (9.50 \times 10^{13} \text{ s}^{-1}) = \underline{6.29 \times 10^{-21} \text{ J}}$$

5. The element with the electron configuration of $[\text{Kr}]5s^24d^15p^1$ is Indium (In).

6. Barium (Ba)

7. Write out the electron configurations for the following elements: (For the exam make sure you know how to do both the complete and the noble gas short cut for electron configurations)

a. Tc Ans: $1s^22s^22p^63s^23p^64s^23d^14p^65s^24d^1$ or $[\text{Kr}]5s^24d^1$

b. Aluminum Ans: $1s^22s^22p^63s^23p^1$ or $[\text{Ne}] 3s^23p^1$

8. Why are Cr and Cu exceptions to writing the electron configurations of elements?

They are exceptions because instead of following the pattern of filling up the d orbitals Cr actually has a half filled s and d orbitals and Cu has a half filled s and completely filled d orbitals. The reason why this occurs is because the electrons in the d-sublevel are more stable with this electron configuration than the expected configuration.

9. a. Ca b. As c. Sn

+ + +

a. 2 . Ca . b. 5 . As . c. 4 . Sn .

+ + +

Periodic Table and Periodic Trends – Chapter 6

1. Compare & contrast how Mendeleev's periodic table was organized to how Moseley organized it.

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OECD Nuclear Energy Agency

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Engineering Separations Unit Operations for Nuclear Processing Reid Peterson, 2019-11-15 Engineering Separations Unit Operations for Nuclear Processing provides insight into the fundamentals of separations in nuclear materials processing not covered in typical texts This book integrates fuel cycle and waste processing into a single coherent approach demonstrating that the principles from one field can and should be applied to the other It provides historical perspectives on nuclear materials processing current assessment and challenges and how past challenges were overcome It also provides understanding of the engineering principles associated with handling nuclear materials This book is aimed at researchers graduate students and professionals in the fields of chemical engineering mechanical engineering nuclear engineering and materials engineering

Sustainable Shipping in a Changing Arctic Lawrence P. Hildebrand, Lawson W. Brigham, Tafsir M. Johansson, 2018-08-11 This volume brings together multiple perspectives on both the changing Arctic environment and the challenges and opportunities it presents for the shipping sector It argues for the adoption of a forward looking agenda that respects the fragile and changing Arctic frontier With the accelerated interest in and potential for new maritime trade routes commercial transportation and

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depressurization without the need to heat water It presents the advantage of utilizing vapor bubble in deaeration applications because the diffusion for degassing happens between the water body and micro vapor bubbles without the need of membrane or packing Instead of relying on massive heating vacuum bubbling focuses on depressurization down to the level of saturated vapor pressure or below to secure vapor bubbles with virtually zero dissolved non condensable gases including oxygen The book considers prospective applications such as extracting high oxygen content air from water for underwater breathing pretreatment of aircraft fuel before being pumped into a fuel tank system and probable desalination applications through massive bubbling combined with low grade renewable energy The book is intended for researchers in thermal fluids heat and mass transfer process engineering and water treatment fields and industry professionals working in power generation plant and process engineering transportation and energy Chemistry and Industry ,1963 Energy Research Abstracts ,1980

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Table of Contents Chemistry Semester Test Review 2012 Answer Key

1. Understanding the eBook Chemistry Semester Test Review 2012 Answer Key
 - The Rise of Digital Reading Chemistry Semester Test Review 2012 Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Semester Test Review 2012 Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Semester Test Review 2012 Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Semester Test Review 2012 Answer Key
 - Personalized Recommendations
 - Chemistry Semester Test Review 2012 Answer Key User Reviews and Ratings
 - Chemistry Semester Test Review 2012 Answer Key and Bestseller Lists

5. Accessing Chemistry Semester Test Review 2012 Answer Key Free and Paid eBooks
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 - Chemistry Semester Test Review 2012 Answer Key Budget-Friendly Options
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 - Chemistry Semester Test Review 2012 Answer Key Compatibility with Devices
 - Chemistry Semester Test Review 2012 Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Semester Test Review 2012 Answer Key
 - Highlighting and Note-Taking Chemistry Semester Test Review 2012 Answer Key
 - Interactive Elements Chemistry Semester Test Review 2012 Answer Key
8. Staying Engaged with Chemistry Semester Test Review 2012 Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Semester Test Review 2012 Answer Key
9. Balancing eBooks and Physical Books Chemistry Semester Test Review 2012 Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Semester Test Review 2012 Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Semester Test Review 2012 Answer Key
 - Setting Reading Goals Chemistry Semester Test Review 2012 Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Semester Test Review 2012 Answer Key
 - Fact-Checking eBook Content of Chemistry Semester Test Review 2012 Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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