

# Chemistry Spring Final Study Guide

### Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
<p><b>Unit 10: Acids &amp; Bases</b></p> <p>Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.</p> <p>Label the y-axis.</p> <p>Label the x-axis.</p> <p>1. A solution of the particles.</p> <p>2. A solution of the particles.</p> <p>3. A solution of the particles.</p> <p>4. A solution of the particles.</p> <p>5. A solution of the particles.</p> <p>6. A solution of the particles.</p> <p>7. A solution of the particles.</p> <p>8. A solution of the particles.</p> <p>9. A solution of the particles.</p> <p>10. A solution of the particles.</p>	<p><b>Unit 10: Acids &amp; Bases</b></p> <p>Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.</p> <p>Label the y-axis.</p> <p>Label the x-axis.</p> <p>1. A solution of the particles.</p> <p>2. A solution of the particles.</p> <p>3. A solution of the particles.</p> <p>4. A solution of the particles.</p> <p>5. A solution of the particles.</p> <p>6. A solution of the particles.</p> <p>7. A solution of the particles.</p> <p>8. A solution of the particles.</p> <p>9. A solution of the particles.</p> <p>10. A solution of the particles.</p>

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Name: \_\_\_\_\_

**Unit 1: Measurement Analysis**

Mark the best answer from those provided to each scenario.

Mark the best answer from those provided to each scenario.

Determine the number of Significant Figures in each of the following numbers.

1. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

2. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

3. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

4. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

5. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

6. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

7. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

8. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

9. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

10. 123.456  $\rightarrow$  \_\_\_\_\_ sig figs

Convert the following into Scientific Notation.

1. 123.456  $\rightarrow$  \_\_\_\_\_

2. 123.456  $\rightarrow$  \_\_\_\_\_

3. 123.456  $\rightarrow$  \_\_\_\_\_

4. 123.456  $\rightarrow$  \_\_\_\_\_

5. 123.456  $\rightarrow$  \_\_\_\_\_

6. 123.456  $\rightarrow$  \_\_\_\_\_

7. 123.456  $\rightarrow$  \_\_\_\_\_

8. 123.456  $\rightarrow$  \_\_\_\_\_

9. 123.456  $\rightarrow$  \_\_\_\_\_

10. 123.456  $\rightarrow$  \_\_\_\_\_

Convert the following into Decimal Notation.

1. 123.456  $\rightarrow$  \_\_\_\_\_

2. 123.456  $\rightarrow$  \_\_\_\_\_

3. 123.456  $\rightarrow$  \_\_\_\_\_

4. 123.456  $\rightarrow$  \_\_\_\_\_

5. 123.456  $\rightarrow$  \_\_\_\_\_

6. 123.456  $\rightarrow$  \_\_\_\_\_

7. 123.456  $\rightarrow$  \_\_\_\_\_

8. 123.456  $\rightarrow$  \_\_\_\_\_

9. 123.456  $\rightarrow$  \_\_\_\_\_

10. 123.456  $\rightarrow$  \_\_\_\_\_

**Unit 2: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 3: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 4: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 5: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 6: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 7: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 8: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 9: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

**Unit 10: Reaction**

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

1.  $H_2O$

2.  $H_2O$

3.  $H_2O$

4.  $H_2O$

5.  $H_2O$

6.  $H_2O$

7.  $H_2O$

8.  $H_2O$

9.  $H_2O$

10.  $H_2O$

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A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the University of Washington text.

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