



# **BUILDING WEALTH: A BEGINNER'S GUIDE TO PERSONAL FINANCE SUCCESS**

The Proven Path to  
Financial Success

# Personal Finance Success Spotlight

**J Rink**



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weak acid this conjugate base is usually a weak base for example sodium acetate  $\text{NaCH}_3\text{CO}_2$  is a salt formed by the reaction of the weak acid acetic acid with the strong base sodium hydroxide

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web summary an arrhenius acid produces  $\text{H}^+$  and an arrhenius base produces  $\text{OH}^-$  in aqueous solutions acids taste sour may sting and neutralize bases bases taste bitter feel slippery and neutralize acids naming acids binary acids contain a single anion  $\text{HX}$  to name hydro anion with  $\text{ICl}_4^-$  end acid

**acids and bases practice problems chemistry steps** - Jul 14 2023

web definition of acids and bases identifying them along with the conjugate acid and conjugate base  $\text{pH}$  of strong and weak acids and bases the  $\text{pH}$  and  $\text{pOH}$  relationship  $K_a$  and  $K_b$   $\text{p}K_a$  and  $K_a$  correlation the acid base properties of salts identifying the acidity of the salt based on the acid and the base it is prepared from and calculating the