

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

**CHAPTER 15** **STUDY GUIDE FOR CONTENT MASTERY**

## Solutions

### Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solvent and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

Study Guide for Content Mastery

Chemistry: Matter and Change • Chapter 15

85

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**CHAPTER 15** **STUDY GUIDE FOR CONTENT MASTERY**

### Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True \_\_\_\_\_ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False \_\_\_\_\_ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False \_\_\_\_\_ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True \_\_\_\_\_ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True \_\_\_\_\_ 18. Additional solute can be dissolved in an unsaturated solution.

True \_\_\_\_\_ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

86 Chemistry: Matter and Change • Chapter 15

Study Guide for Content Mastery

# Content Mastery Study Guide Answers

**Hector M. Adams**



## **Content Mastery Study Guide Answers:**

*Study Guide to DSM-5-TR®* Laura Weiss Roberts, Alan K. Louie, Matthew L. Edwards, 2024-12-12 The ultimate companion volume to DSM 5 TR the Study Guide is designed to help clinical learners teachers and practitioners in psychiatry psychology and social work understand and apply diagnostic criteria Medical students residents and fellows and trainees throughout the clinical neurosciences will find this text to be invaluable in developing their diagnostic abilities Readers learn key clinical content through concise chapters and can assess their knowledge with more than 100 multiple choice questions The book is organized into three sections Part I delves into foundational concepts of psychiatric diagnoses and explores how diagnostic assessment fits into the process of working therapeutically with patients In this new edition Part I includes a new chapter examining structural and cultural considerations in arriving at a diagnosis Part II focuses on the diagnostic classes within DSM 5 TR and features clinical case vignettes with real life complexity relevant to psychiatric diagnoses diagnostic pearls that can be applied to daily clinical practice and a self assessment section with key concepts questions for further discussion patient cases and Q As Part III includes more than 100 multiple choice questions with an answer key These questions cover a range of diagnoses and patient presentations from across DSM 5 TR helping readers review their learning and assess their skills Taking a patient centered approach that complements the more disorder centered organization of DSM 5 TR this Study Guide provides context for DSM 5 TR s diagnoses while offering an array of interesting and memorable cases Rich in thought provoking detail and designed for repeated use this is an indispensable guide for learners educators and clinicians

**Statistics for Nursing Research - E-Book** Susan K. Grove, Daisha J. CIPHER, 2024-01-20 Get the review and practice you need to critically appraise the statistical methods used in published research Statistics for Nursing Research 4th Edition helps you understand and interpret statistical methods sampling and measurement techniques and statistical analysis techniques This unique and practical workbook is divided into two parts to differentiate between basic and advanced statistical methods with dozens of hands on exercises that will help you practice working with the statistical methods needed for building an evidence based practice This workbook is an indispensable resource for any nursing student or practicing nurse wishing to gain competency in conducting statistical analyses Comprehensive coverage of basic statistical methods includes extensive practice with the most common techniques of sampling measurement and statistical analysis seen in nursing and health sciences literature Literature based approach uses key excerpts from published studies to reinforce learning through practical application 38 sampling measurement and statistical analysis exercises provide a practical review of both basic and advanced statistical techniques Study Questions in each chapter help you examine a statistical technique and apply concepts to an actual literature appraisal Online submission of answers for Questions for Further Application allows you to complete exercises in a web based format and then submit completed exercises to your instructor via email for grading Consistent format for all chapters enhances learning and enables quick review NEW More than half of the research

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## **Table of Contents Content Mastery Study Guide Answers**

1. Understanding the eBook Content Mastery Study Guide Answers
  - The Rise of Digital Reading Content Mastery Study Guide Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Content Mastery Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Content Mastery Study Guide Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Content Mastery Study Guide Answers

- Personalized Recommendations
- Content Mastery Study Guide Answers User Reviews and Ratings
- Content Mastery Study Guide Answers and Bestseller Lists
- 5. Accessing Content Mastery Study Guide Answers Free and Paid eBooks
  - Content Mastery Study Guide Answers Public Domain eBooks
  - Content Mastery Study Guide Answers eBook Subscription Services
  - Content Mastery Study Guide Answers Budget-Friendly Options
- 6. Navigating Content Mastery Study Guide Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Content Mastery Study Guide Answers Compatibility with Devices
  - Content Mastery Study Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Content Mastery Study Guide Answers
  - Highlighting and Note-Taking Content Mastery Study Guide Answers
  - Interactive Elements Content Mastery Study Guide Answers
- 8. Staying Engaged with Content Mastery Study Guide Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Content Mastery Study Guide Answers
- 9. Balancing eBooks and Physical Books Content Mastery Study Guide Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Content Mastery Study Guide Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Content Mastery Study Guide Answers
  - Setting Reading Goals Content Mastery Study Guide Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Content Mastery Study Guide Answers



- Fact-Checking eBook Content of Content Mastery Study Guide Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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