

Chemistry: Ionic & Covalent Bonds Unit Study Guide

Covalent Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Covalent Compounds.

- Bond (ion)
- (Large / Small) Electronegativity Difference
- (Same / Different) Elements
- (High / Low) Melting & Boiling Point
- (Conducts / Does not Conduct) Electricity in Solution
- (Solid / Liquid / Gas) at Room Temperature

2. Write the prefix following number

Atoms	Prefix
1	
2	
3	
4	
5	

4. What are the two

5. Complete the chart

Formula	Name
BF_3	
	Fe
S_2I_2	
	C

6. What does VSEPR

Name: _____ Date: _____ Period: _____

Ionic Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Ionic Compounds.

- Bond between a (Metal / Nonmetal) and (Metal / Nonmetal)
- (Large / Small) Electronegativity Difference
- The (Metals / Nonmetals) form Cations, which (Gain / Lose) Electrons
- The (Metals / Nonmetals) form Anions, which (Gain / Lose) Electrons
- (High / Low) Melting & Boiling Point
- (Conducts / Does not Conduct) Electricity in Solution
- (Solid / Liquid / Gas) at Room Temperature

2. What charge would the following periodic table families have in a chemical bond?

Family	Charge
alkali metals	
alkali earth	
halogens	
transition	
nitrogen	
oxygen	

3. Write the roman numeral that is represented by the following numbers.

Roman Numeral	Number
	1
	2
	3
	4
	5
	6

4. Complete the chart below by filling in the missing boxes.

CATION	ANION	FORMULA	NAME
Mg^{2+}	Br^{-}		
		CuO	
			Iron (III) chloride
		$(NH_4)_3PO_4$	

Covalent Bonds Study Guide

**John Krenos, Joseph Potenza, Loretta
Jones, Lynn Koplitz, Thomas Spence**



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points from the text presents problems and hints to assist the student in solving them Solutions to selected problems in the end of chapter problem sets are included By Kenneth Hughes of the University of Wisconsin Madison **Study Guide with Solutions to Accompany Burgoyne, A Study Course in Organic Chemistry** N. Sue Barnes, 1979

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