

Chemistry — Unit 8 Test

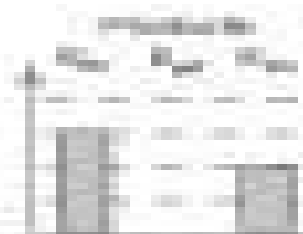
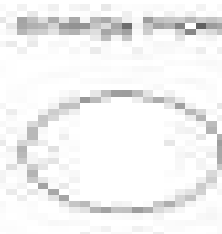
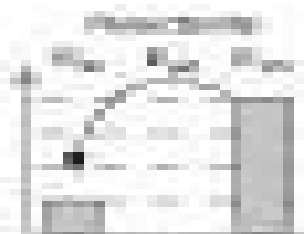
1. In this ideal gas law _____ represents number of gas _____
 n P V T

2. The activation energy for a reaction _____
 a. The activation energy decreases as the products _____
 b. The products have more energy than the reactants _____
 c. The kinetic energy of the system decreases _____
 d. All of the positions is positive _____

3. Consider the energy level shown _____
 as right _____

Which of the following _____
 state about the _____
 reaction is FALSE _____

- a. The reaction is exothermic _____
 b. The products have less _____
 energy than the reactants _____
 c. The temperature of the _____
 system decreases _____
 d. The energy level would appear _____ right side of the equation _____



Match volume and gas solubility

4. The three gas solubility laws state that _____
 a. The three gas solubility laws state that _____
 b. The three gas solubility laws state that _____
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 z. The three gas solubility laws state that _____

6. When potassium chlorate, $KClO_3$, is decomposed, oxygen gas and potassium chloride _____
 are produced. What volume of oxygen gas at STP is produced when 2.75 g of $KClO_3$ _____
 decomposes? _____
 Answer: _____

Chemistry Unit 9 Test

Jianjun Gao

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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