

Intro. to Chemical Reactions

Electrolyte Reactions

1. Write down formulas for the following compounds:

a) Barium hydroxide barium hydroxide

b) calcium phosphate calcium phosphate

c) magnesium phosphate magnesium phosphate

d) lithium iodide lithium iodide

2. Write the formulas for the following acids:

a) H₂SO₄ hydrogen sulfate

b) H₃PO₄ hydrogen phosphate

c) H₂CO₃ hydrogen carbonate

d) HClO chlorous acid

Electrolyte Reactions: Physical Theory Review

There is often a chemical reaction that does not go on...



There are reasons that you can come up with when this goes on...

Common/less reactive reactants with **insoluble** and **unstable** and **gaseous** products

There is also no electron movement between the reactants and the products but there is no charge transfer...

Corrosive/Unstable Compounds

1. Given the following reaction, write the equation that goes along with it...



Corrosive/Unstable products with either water and acidic or basic nature and a large amount of energy released

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