

# Matter—Properties and Changes

## Section 3.1 Properties of Matter

In your textbook, read about *physical properties* and *chemical properties* of matter.

Use each of the terms below just once to complete the passage.

chemical  
density

mass  
properties

physical  
substance

Matter is anything with (1) \_\_\_\_\_ and volume. A (2) \_\_\_\_\_ is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) \_\_\_\_\_ that can be observed. Substances have both physical and chemical properties. (4) \_\_\_\_\_ properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) \_\_\_\_\_ are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) \_\_\_\_\_ properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

\_\_\_\_\_ 7. Chemical formula  $\text{H}_2\text{O}$   
\_\_\_\_\_ 8. Forms green carbonate when exposed to moist air  
\_\_\_\_\_ 9. Remains unchanged when in the presence of nitrogen  
\_\_\_\_\_ 10. Colorless  
\_\_\_\_\_ 11. Solid at normal temperatures and pressures  
\_\_\_\_\_ 12. Ability to combine with another substance  
\_\_\_\_\_ 13. Melting point  
\_\_\_\_\_ 14. Liquid at normal temperatures and pressures  
\_\_\_\_\_ 15. Boiling point is 100°C  
\_\_\_\_\_ 16. Conducts electricity  
\_\_\_\_\_ 17. Density is  $\frac{1\text{g}}{\text{cm}^3}$

# Chemistry Study Guide Packet Answer

**L Darling-Hammond**

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