

Unit 6 Study Guide

TOPIC 1: REVIEW

Aq (aqueous) - dissolved in water

Solution: Homogenous mixture; solutes dissolved in solvents

Solute: Dissolved particles in a solution (i.e. NaCl)

Solvent: The dissolved medium in a solution



Saturated: Solution containing max amount of solute for a given amount of solvent at a constant temperature and pressure

TOPIC 2: SOLUBILITY

- Water molecule has $\ominus$ and $\oplus$ end
- negative attached to cation
- positive attached to anion



Ion solution: water surrounding ions

- When certain IONIC COMPOUNDS (compounds w/ 1+ atoms) are added to water, they will break apart / DISSOCIATE



Solubility rules:

Chemistry Unit 6 Study Guide

Tao Wei

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Embracing the Tune of Phrase: An Psychological Symphony within **Chemistry Unit 6 Study Guide**

In a world consumed by displays and the ceaseless chatter of quick interaction, the melodic beauty and psychological symphony created by the prepared word usually fade into the background, eclipsed by the persistent noise and disturbances that permeate our lives. But, situated within the pages of **Chemistry Unit 6 Study Guide** a stunning fictional value overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Crafted by a wonderful musician of language, this interesting masterpiece conducts viewers on an emotional trip, well unraveling the concealed melodies and profound impact resonating within each cautiously constructed phrase. Within the depths of this emotional analysis, we will explore the book is key harmonies, analyze their enthralling writing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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