



Atomic Structure and Nuclear Chemistry Study Guide

Topics Covered:

Subatomic particles and the nucleus

Isotopes and Ions

Atomic mass and atomic number

Average atomic mass and isotopic abundance

Nuclear stability

Ionizing radiation

Alpha decay and beta decay

Transmutation equations

Fission vs. fusion

Mass spectroscopy

Half-life

Practice Questions: [Click here for an answer key](#)

1. What does it mean for an atom to be neutral? What does it mean for different subatomic particles play here?
2. What is the relationship between the atomic number/atomic mass and the number of protons and neutrons?

Chemistry Study Guide Nuclear Chemistry

**Gregory Choppin,Jan-Olov Liljenzin,Jan
Rydberg,Christian Ekberg**

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Radiochemistry and Nuclear Chemistry Gregory Choppin, Jan-Olov Liljenzén, Jan Rydberg, Christian Ekberg, 2013-09-05

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