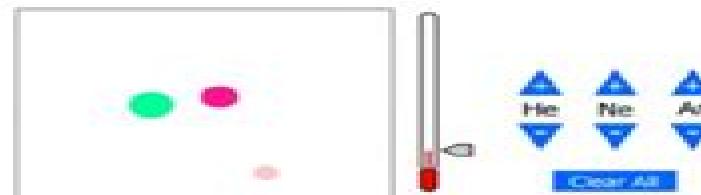


Chemthink – Behavior of Gases

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1. Use the + button to add an atom of helium, neon, and argon to the container of gases. Rank each based on atom size (from smallest to largest).
2. Rank each based on speed (fastest to slowest).
3. How does adding heat affect the behavior of the atoms?
4. Remove all of the atoms except for helium. Add 4-5 more helium atoms. Do they all move at the same speed?
5. Give the definition and equation for pressure.
 - a. Equation:
 - b. Definition:
6. What does psi stand for?
 - a. What is normal atmospheric pressure?
7. What causes the force of pressure?
8. Each second, a square inch of your hand is hit by molecules an average of _____ times. (Write your answer in scientific notation!)
 - a. Each molecule has an average speed of about _____.
9. Slide the temperature gauge up and down and click the "take a reading" button at several points until you can see what the graph should look like. Draw it.



Chemthink Behavior Of Gases Answers

Michal Rosen-Zvi

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Chemistry and Our Universe , In the first of two lectures on the properties of gases review the basic equations that describe their behavior Learn the history of Boyle s law Gay Lussac s law Charles s law and Avogadro s law Then use these four expressions to derive the celebrated ideal gas law **Nonideal Behavior of Gases and Their Mixtures** Abdus Satter,1963

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