

CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

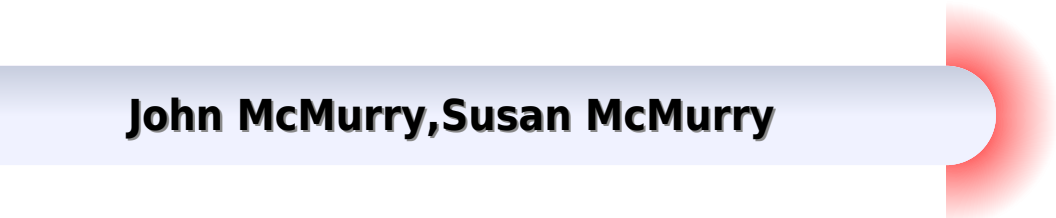
- | | |
|----------------|---|
| <u> false </u> | 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed. |
| <u> true </u> | 2. The most abundant substance in a colloid is the dispersion medium. |
| <u> false </u> | 3. A colloid can be separated by filtration. |
| <u> true </u> | 4. A solid emulsion consists of a liquid dispersed in a solid. |
| <u> true </u> | 5. Whipped cream is an example of a foam. |
| <u> false </u> | 6. In an aerosol, the dispersing medium is a liquid. |
| <u> true </u> | 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles. |
| <u> true </u> | 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces. |
| <u> false </u> | 9. Stirring an electrolyte into a colloid stabilizes the colloid. |
| <u> true </u> | 10. Colloids demonstrate the Tyndall effect. |

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter		✓	✓
12. Between 1 nm and 1000 nm in diameter	✓	✓	✓
13. More than 1000 nm in diameter	✓	✓	
14. Settle out if undisturbed			
15. Pass through standard filter paper	✓		
16. Lower vapor pressure			✓
17. Scatter light	✓		

Chemistry Study Guide For Mixtures And Solutions

John McMurry, Susan McMurry



Chemistry Study Guide For Mixtures And Solutions:

Student Study Guide to accompany Chemistry Martin Silberberg,Libby Weberg,2005-01-06 **Organic Chemistry Study Guide with Solutions Manual** Neil E. Schore,2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer *Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition* Susan McMurry,2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Price winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines **General Chemistry Study Guide Sixth Edition** Darrell D. Ebbing,1999 **Study Guide/Selected Solutions Manual** Julie R. Frentrop,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book *Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry* John McMurry,Susan McMurry,1986 Study Guide for E. Russell Hardwick's Introduction to Chemistry Joan Bouillon,1984 *Solutions Manual and Study Guide to Accompany Introduction to Organic Chemistry, 4th Ed* Paul A. Bartlett,Judith G. Koch,1992 *Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry* McMurry,Susan McMurry,1990 **Study Guide & Solutions Manual to Accompany Organic Chemistry** Mary H. Bailey,1998 Study Guide for Ausubel/Robinson School Learning Floyd G. Robinson,1970 *Resources in Education* ,1984 *Essential Concepts of Chemistry Study Guide* James R. Braun,Sherman,1999 *Study Guide for General Chemistry for Colleges* Herman Thompson Briscoe,1936 *Differentiated Study Guide High School Chemistry* Carl David Morneweck,Chester Arthur Buckner,Gale R. Kirschner,1935 *Study guide to accompany Drew H. Wolfe: General, organic and biological chemistry* Daniel C. Pantaleo,1986 **Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 **General Chemistry** Darrell D. Ebbing,Larry K. Krannich,Joan I. Senyk,1996 Biology Glencoe/McGraw-Hill,1994-07 *An Introductory Guide to EC Competition Law and Practice* Valentine Korah,1994

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