

ICSE Class 10 Chemistry Question Paper Solution 2016

CHEMISTRY SCIENCE Paper – 2

Question 1

(a) Fill in the blanks with the choices given in brackets. [5]

- Metals are good _____ (oxidizing agents / reducing agents) because they are electron _____ (acceptors/ donors).
- Electrovalent compounds have _____ (high / low) melting points.
- Higher the pH value of a solution, the more _____ (acidic / alkaline) it is.
- _____ ($AgCl$ / $PbCl_2$), a white precipitate is soluble in excess NH_4OH .
- Conversion of ethene to ethane is an example of _____ (hydration / hydrogenation).

(b) Choose the *correct answer* from the options given below: [5]

- An element with the atomic number 19 will most likely combine chemically with the element whose atomic number is:
 - 17
 - 11
 - 18
 - 20
- The ratio between the number of molecules in 2g of hydrogen and 32g of oxygen is:
 - 1:2
 - 1:0.01
 - 1:1
 - 0.01:1 [Given that H=1, O=16]
- The two main metals in *Bronze* are:
 - Copper and zinc
 - Copper and lead
 - Copper and nickel
 - Copper and tin
- The particles present in *strong electrolytes* are:
 - only molecules
 - mainly ions

Chemistry Test Questions Fill In The Blank

Albert A Gayle

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