

Chemistry Study Guide

Periodic Table of the Elements

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| ³ ₃ Li | ⁴ ₄ Be | | | | | | | | | | | | | | | | | ⁵ ₅ B | ⁶ ₆ C | ⁷ ₇ N | ⁸ ₈ O | ⁹ ₉ F | ¹⁰ ₁₀ Ne | | | | | | | | | | | | | | | | | | | | | | | | |
| ¹¹ ₁₁ Na | ¹² ₁₂ Mg | | | | | | | | | | | | | | | | | ¹³ ₁₃ Al | ¹⁴ ₁₄ Si | ¹⁵ ₁₅ P | ¹⁶ ₁₆ S | ¹⁷ ₁₇ Cl | ¹⁸ ₁₈ Ar | | | | | | | | | | | | | | | | | | | | | | | | |
| ¹⁹ ₁₉ K | ²⁰ ₂₀ Ca | ²¹ ₂₁ Sc | ²² ₂₂ Ti | ²³ ₂₃ V | ²⁴ ₂₄ Cr | ²⁵ ₂₅ Mn | ²⁶ ₂₆ Fe | ²⁷ ₂₇ Co | ²⁸ ₂₈ Ni | ²⁹ ₂₉ Cu | ³⁰ ₃₀ Zn | ³¹ ₃₁ Ga | ³² ₃₂ Ge | ³³ ₃₃ As | ³⁴ ₃₄ Se | ³⁵ ₃₅ Br | ³⁶ ₃₆ Kr | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| ³⁷ ₃₇ Rb | ³⁸ ₃₈ Sr | ³⁹ ₃₉ Y | ⁴⁰ ₄₀ Zr | ⁴¹ ₄₁ Nb | ⁴² ₄₂ Mo | ⁴³ ₄₃ Tc | ⁴⁴ ₄₄ Ru | ⁴⁵ ₄₅ Rh | ⁴⁶ ₄₆ Pd | ⁴⁷ ₄₇ Ag | ⁴⁸ ₄₈ Cd | ⁴⁹ ₄₉ In | ⁵⁰ ₅₀ Sn | ⁵¹ ₅₁ Sb | ⁵² ₅₂ Te | ⁵³ ₅₃ I | ⁵⁴ ₅₄ Xe | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| ⁵⁵ ₅₅ Cs | ⁵⁶ ₅₆ Ba | | ⁵⁷ ₅₇ Hf | ⁵⁸ ₅₈ Ta | ⁵⁹ ₅₉ W | ⁶⁰ ₆₀ Re | ⁶¹ ₆₁ Os | ⁶² ₆₂ Ir | ⁶³ ₆₃ Pt | ⁶⁴ ₆₄ Au | ⁶⁵ ₆₅ Hg | ⁶⁶ ₆₆ Tl | ⁶⁷ ₆₇ Pb | ⁶⁸ ₆₈ Bi | ⁶⁹ ₆₉ Po | ⁷⁰ ₇₀ At | ⁷¹ ₇₁ Rn | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| ⁷² ₇₂ Fr | ⁷³ ₇₃ Ra | | ⁷⁴ ₇₄ Rf | ⁷⁵ ₇₅ Db | ⁷⁶ ₇₆ Sg | ⁷⁷ ₇₇ Bh | ⁷⁸ ₇₈ Hs | ⁷⁹ ₇₉ Mt | ⁸⁰ ₈₀ Ds | ⁸¹ ₈₁ Rg | ⁸² ₈₂ Cn | ⁸³ ₈₃ Uut | ⁸⁴ ₈₄ Fl | ⁸⁵ ₈₅ Uup | ⁸⁶ ₈₆ Lv | ⁸⁷ ₈₇ Uus | ⁸⁸ ₈₈ Uuo | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location |
|----------|--------|------------|--------|-------------|
| Proton | p^+ | 1 | +1 | Nucleus |
| Neutron | n | 1 | 0 | Nucleus |
| Electron | e^- | ~ 0 | -1 | Orbit/Shell |

Chemistry Study Guide Pearson

Karen C. Timberlake



Chemistry Study Guide Pearson:

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