

## Chemistry — Unit 8 Test

1. The following graph shows the relationship between the number of atoms and the mass of an element.

- a. The two axes for this graph are:
- b. The following statement is true concerning this graph:
  - i. The positive slope means increasing the mass does not increase the number of atoms.
  - ii. The negative slope means increasing the mass does not increase the number of atoms.
  - iii. The horizontal slope means the number of atoms does not change.
  - iv. The vertical slope means the number of atoms does not change.

2. The following statement is true concerning these elements:

- a. Elements of the same block have similar relative atomic masses.
- b. The relative atomic mass of an element is the same as its atomic number.
- c. The positive slope of the graph below shows the relationship between the mass of an element and the number of atoms.
- d. The negative slope of the graph below shows the relationship between the mass of an element and the number of atoms.



### Relative reactivities of metals against dilute hydrochloric acid

3. The following table shows the relative reactivities of metals against dilute hydrochloric acid. The greater the value, the greater the reactivity of the metal against dilute hydrochloric acid.

Metals	Relative reactivity
Aluminum	0.1
Iron	0.2
Magnesium	0.3
Aluminum	0.4
Iron	0.5
Magnesium	0.6
Aluminum	0.7
Iron	0.8
Magnesium	0.9
Aluminum	1.0
Iron	1.1
Magnesium	1.2

4. The following table shows the relative reactivities of metals against dilute sulfuric acid. The greater the value, the greater the reactivity of the metal against dilute sulfuric acid.

Metals	Relative reactivity
Aluminum	0.1
Iron	0.2
Magnesium	0.3
Aluminum	0.4
Iron	0.5
Magnesium	0.6
Aluminum	0.7
Iron	0.8
Magnesium	0.9
Aluminum	1.0
Iron	1.1
Magnesium	1.2

# Chemistry Unit 9 Test

**Annelies Wilder-Smith**

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