

Intro to Chemistry Final Exam Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question.

1. The Fahrenheit temperature scale is defined by which of the following temperatures?
 - a. Water boils at 100 degrees and freezes at 32 degrees.
 - b. Water boils at 212 degrees and freezes at 32 degrees.
 - c. Water evaporates at 212 degrees and freezes at 0 degrees.
 - d. Liquid water turns to a gas at 100 degrees and to a solid at 0 degrees.
2. A cold-blooded reptile basks on a warm rock in the sun. Its body is warmed by
 - a. radiation.
 - b. conduction.
 - c. convection.
 - d. Both (a) and (b)
3. Temperature is a measure of the average _____ energy of an object's particles.
 - a. mechanical
 - b. kinetic
 - c. potential
 - d. light
4. The temperature at which all molecular motion stops is
 - a. 0°C.
 - b. 0°F.
 - c. 0 K.
 - d. All of the above
5. Convert 500°F to degrees Celsius.
 - a. 260°C
 - b. 290°C
 - c. 84.2°C
 - d. 95.8°C
6. Convert 300 K to the Celsius scale.
 - a. 0°C
 - b. 27°C
 - c. 73°C
 - d. 573°C
7. Convert 458 K to the equivalent Fahrenheit temperature.
 - a. 185°F
 - b. 237°F
 - c. 365°F
 - d. 731°F
8. Which of the following temperatures is impossible to measure?
 - a. -85°F
 - b. -50°C
 - c. -20 K
 - d. 545°F
9. What is -173°C on the Kelvin scale?
 - a. 76 K
 - b. 89 K
 - c. 98 K
 - d. 448 K
10. The principles of atomic theory recognized today were conceived by
 - a. Avogadro.
 - b. Bohr.
 - c. Dalton.
 - d. Rutherford.
11. The total number of protons and neutrons in the nucleus of an atom is its
 - a. atomic number.
 - b. Avogadro number.
 - c. mass number.
 - d. average atomic mass.
12. All atoms of the same element have the same
 - a. atomic mass.
 - b. number of neutrons.
 - c. mass number.
 - d. atomic number.
13. Atoms of the same element can differ in
 - a. chemical properties.
 - b. mass number.
 - c. atomic number.
 - d. number of protons and electrons.
14. Chlorine has atomic number 17 and mass number 35. It has
 - a. 17 protons, 17 electrons, and 18 neutrons.
 - b. 35 protons, 35 electrons, and 17 neutrons.
 - c. 17 protons, 17 electrons, and 52 neutrons.
 - d. 18 protons, 18 electrons, and 17 neutrons.
15. Phosphorus-33 (atomic number 15) contains
 - a. 33 protons.
 - b. 18 neutrons.
 - c. 33 neutrons.
 - d. 18 protons.
16. Neon-22 contains 12 neutrons. It also contains
 - a. 12 protons.
 - b. 22 protons.
 - c. 22 electrons.
 - d. 10 protons.
17. Valence electrons determine an atom's

Chemistry Study Guide Multiple Choice Answers

Arshad Iqbal

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book *Team-Based Learning* Larry K. Michaelsen, Arletta Bauman Knight, L. Dee Fink, 2023-07-03 This book describes team based learning TBL an unusually powerful and versatile teaching strategy that enables teachers to take small group learning to a whole new level of effectiveness. It is the only pedagogical use of small groups that is based on a recognition of the critical difference between groups and teams and intentionally employs specific procedures to transform newly formed groups into high performance learning teams. This book is a complete guide to implementing TBL in a way that will promote the deep learning all teachers strive for. This is a teaching strategy that promotes critical thinking, collaboration, mastery of discipline knowledge and the ability to apply it. Part I covers the basics beginning with an analysis of the relative merits and limitations of small groups and teams. It then sets out the processes with much practical advice for transforming small groups into cohesive teams for creating effective assignments and thinking through the implications of team based learning. In Part II teachers from disciplines as varied as accounting, biology, business, ecology, chemistry, health education and law describe their use of team based learning. They also demonstrate how this teaching strategy can be applied equally effectively in environments such as large classes, mixed traditional and online classes, and with highly diverse student populations. Part III offers a synopsis of the major lessons to be learned from the experiences of the teachers who have used TBL as described in Part II. For teachers contemplating the use of TBL this section provides answers to key questions e.g. whether to use team based learning, what it takes to make it work effectively and what benefits one can expect from it for the teacher as well as for the learners. The appendices answer frequently asked questions, include useful forms and exercises, and offer advice on peer evaluations and grading. A related Web site that allows readers to continue the conversation, view video material, access indexed descriptions of applications in various disciplines and post questions further enriches the book. The editors claim that team based instruction can transform the quality of student learning, is fully supported by the empirical evidence and examples they present. An important book for all teachers in higher education. [Professional Careers Sourcebook](#) Sara T. Bernstein, Kathleen M. Savage, 1995-09 Where to find help planning careers that require college or technical degrees

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