

3 • Organic Chemistry**UNIT TEST • PRACTICE**

- You may use a pencil, eraser, and scientific calculator to complete the test. No other resources are allowed.
- You have fifty (50) minutes to complete this test, unless other arrangements have been made.
- Please transfer your answers for Matching and Multiple Choice questions onto the Answer Document. Work for the Matching and Multiple Choice questions will not be graded. However, sufficient and appropriate work must be shown for the Free Response questions in order to receive full credit.

SECTION 1: MATCHING. Use the following answer choices that best match each description. Answer choices may be used once, more than once, or not at all. (1 point each)

Questions 1-6:

- (A) Alkanes only
- (B) Alkenes only
- (C) Alkynes only
- (D) Both alkenes and alkynes

1. Have a molecular formula C_4H_8 .
2. Have *cis-trans*- isomers.
3. Can rotate around C–C bond.
4. Have a triple bond in the main carbon chain.
5. Have an H–C–C bond angle of 120° .
6. C_4H_6 is an example.
7. Are considered "saturated" hydrocarbons.

Questions 7-12:

- (A) Ketone
- (B) Amine
- (C) Carboxylic Acid
- (D) Ether

8. Has a carbon bonded to a nitrogen.
9. Has a sour taste.
10. Acetone, or nail polish remover, is an example.
11. Has a carbon that is bonded to oxygen only with a double bond.
12. Has an oxygen single bonded to two carbons.
13. Vinegar is an example.

SECTION 2: MULTIPLE CHOICE. Select the best answer choice for each question. (2 points each)

14. Which of the following elements has a bonding capacity of 2?
 - (A) Oxygen
 - (B) Hydrogen
 - (C) Carbon
 - (D) Chlorine

15. Structural formulas differ from molecular formulas in that they show
 - (A) The arrangement of atoms.
 - (B) The bond angles in the molecule.
 - (C) The elements that make up the compound.
 - (D) The number of each element present.

Chemistry Unit Test Practice

Earl Rouse Glenn, Louis E. Welton

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