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pre laboratory questions procedure answer on a separate sheet of paper 1

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beans mass of beans of this kind atomic mass of this bean isotope abundanc e kidney bean 32 36 1 122 black eye bean 98 23
0 25 white bean 30 16 0 50 calculation of atomic mass of beanium 36 32 1 amu

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web calculate theaverageatomic massofbeaniumtobeplaced ontheperiodic table beanium isotope labclass set data totalmass
atoms total mass isotope 1 atoms 1 avg mass 1 total mass isotope 2 atoms 2 avg mass 2 total mass isotope 3 atoms 3 avg
mass 3 total mass ofallatoms total of atoms

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web isotopes and weighted average atomic mass amu atomic mass unit mass of one proton mass of one neutron a sample of
cesium cs has the following abundance cs 132 20 0 cs 133 75 3 cs 134 4 7 weighted average atomic mass 132×0.200 133×0.752 134×0.047 132 85 amu

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element are isotopes of each other ordinary beans are a lot bigger than atoms but perhaps they can give you one or two clues
about isotopes we have talked in class about isotopes atoms of the same element that differ in

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how you would determine the average atomic mass of an element section 2 data table don t forget calibration and units total
all beans 100 beanium isotope blackium beanium isotope brownium beanium

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web beanium isotope b brownium beanium isotope c whiteium of atoms beans mass beans in cup mass empty cup mass of atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c x 100 section 3 calculate the average atomic mass of beanium

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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web good luck saftey 1 do not eat the beans 2 do not throw the beans 3 do not crush the beans don t mess with the beans pre lab questions 1 there are many beans in your container each bean represents an 2 there are three kinds of beans in your container each kind represents a different