

Chapter 12 – Stoichiometry Study Guide

The following example problems exhibit the types of calculations you will be expected to perform on tomorrow's test.

- Described a reaction in terms of particles.

Ex. 1 Describe the following reaction in terms of molecules.



- “Note: sometimes you will need to include other descriptors such as atoms and formula units.”

- Described a reaction in terms of moles.

Ex. 2 Describe the following reaction in terms of moles.



- How do you determine the mass of the products for a balanced reaction?

Ex. 3 What is the mass of the product formed in the following equation?



- How do you determine the volume of the reactants or products of a balanced reaction?

Ex. 4 What is the volume of the products and the reactants of the following equation?



- How do you show how a balanced reaction follows the Law of Conservation of Mass by showing that the mass on both sides of a reaction are equal?

Ex. 5 Show how the mass of the reactants equals the mass of the products for the following equation.

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**Denise Guinn, Rebecca Brewer, Rachel
C. Lum**

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