

Chapter Test B

A. Matching

Match each item in Column A with the correct description in Column B. Write the letter of the correct description in the blank.

Column A

1. An unknown mass dissolved in a known volume of solvent for solution formation and analysis
2. An unknown volume of gas produced from a known mass of reactant
3. The relationship of quantities in chemical equations
4. The ratio of the actual yield to the theoretical yield expressed as a percent
5. The unknown mass of a product or reactant in a chemical reaction
6. An unknown amount of substance that reacts with a known amount of substance

Column B

- a. percent yield
- b. limiting reagent
- c. theoretical yield
- d. stoichiometry
- e. percent yield
- f. percent reagent

B. Multiple Choice

Choose the best answer for each item from the list.

1. Which of the following is correct?
a. moles of substance
b. grams of substance
c. percent of substance
d. grams per substance
e. moles per substance
f. mass-moles per substance per substance
2. Which of the following is correct?
a. 1 mole of H_2O is 18 grams
b. 1 mole of H_2O is 18 moles
c. 1 mole of H_2O is 18 moles
d. 1 mole of H_2O is 18 moles
e. 1 mole of H_2O is 18 moles
f. 1 mole of H_2O is 18 moles
3. Which of the following is correct?
a. 1 mole of H_2O is 18 grams
b. 1 mole of H_2O is 18 moles
c. 1 mole of H_2O is 18 moles
d. 1 mole of H_2O is 18 moles
e. 1 mole of H_2O is 18 moles
f. 1 mole of H_2O is 18 moles

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Stoichiometry Key

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