

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY**Solutions****Section 15.1 What are solutions?**

In your textbook, read about the characteristic of solutions.

Underline the terms below just now to complete the passage.

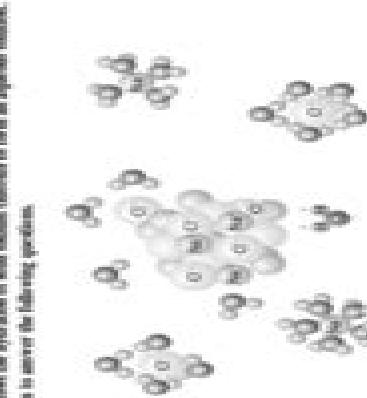
solvent	liquid	solvent	solvent
insoluble	solid	solid	solid

A(n) **10. solution** of sugar is dissolved in water.The oxygen atom in the **11. water** molecule is the **12. oxygen** atom.

13. solute because we put the **14. solute** in a solution, or **15. in an 16. solution**. A solution that does not dissolve is a **17. insoluble** (b) **18. liquid** solution of the most common type of solution. If one liquid is soluble in another liquid, such as water and oil, water is the **19. solute** in the liquid in **20. (b) immiscible**. The liquid in **21. immiscible**.

Read about solubility in aqueous solution in your textbook.

Underline the questions to answer the following questions.



The diagram shows the hydration of solid sodium chloride to form an aqueous solution.

Underline the diagram to answer the following questions.

1. In water, a solute is a substance that is dissolved in water. What is a solute?
Solution is the process of **surrounding** solute particles with solvent particles to form a solution.

2. True _____ False _____ In a saturated solution, solute and crystallization are in equilibrium.

3. True _____ False _____ Additional solute can be dissolved in an unsaturated solution.

4. True _____ False _____ The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Section 15.1 (cont.)

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Chapter 12 Study Guide For Content Mastery Section 122

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