

**CHAPTER 14****STUDY GUIDE**

## Mixtures and Solutions

### Section 14.1 Heterogeneous and Homogeneous Mixtures

In your notebook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

_____	1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.
_____	2. The most abundant substance in a colloid is the dispersion medium.
_____	3. A colloid can be separated by filtration.
_____	4. A solid emulsion consists of a liquid dispersed in a solid.
_____	5. Whipped cream is an example of a foam.
_____	6. In an aerosol, the dispersing medium is a liquid.
_____	7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
_____	8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
_____	9. Stirring an electrolyte into a colloid stabilizes the colloid.
_____	10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter		<input checked="" type="checkbox"/>	
12. Between 1 nm and 1000 nm in diameter	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>
13. More than 1000 nm in diameter	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>	
14. Settle out if undisturbed			
15. Pass through standard filter paper	<input checked="" type="checkbox"/>		
16. Lower vapor pressure	<input checked="" type="checkbox"/>		<input checked="" type="checkbox"/>
17. Scatter light	<input checked="" type="checkbox"/>		

# Chapter 14 Study Guide Chemistry Mixtures Solutions

**Barbara Herlihy**

## **Chapter 14 Study Guide Chemistry Mixtures Solutions:**

*Chemistry, Student Study Guide* James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon are buckyballs, graphite, and diamond. They are illustrated at the left, as is the molecule methane CH<sub>4</sub> from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement.

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