



Electrochemical Cells, continued

Half-Cell, continued

- Both oxidation and reduction must occur in an electrochemical reaction.
 - The two half-cells taken together make an electrochemical cell.

The Complete Cell

- An electrochemical cell may be represented by the following notation:

anode electrode|anode solution||cathode solution|cathode electrode

- The double line represents the salt bridge, or the porous barrier.



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