



Electrochemical Cells, *continued* Half-Cell, *continued*

- Both oxidation and reduction must occur in an electrochemical reaction.
 - The two half-cells taken together make an electrochemical cell.

The Complete Cell

- An electrochemical cell may be represented by the following notation:

anode electrode|anode solution||cathode solution|cathode electrode

- The double line represents the salt bridge, or the porous barrier.



Chapter 20 Review Electrochemistry Section 1

SILBERBERG

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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