

NUCLEAR CHEMISTRY

BALANCING NUCLEAR REACTIONS WORKSHEET

Predict the missing product or reactant in the following nuclear reactions. Determine the type of nuclear reaction (α emission, β emission, γ emission, positron emission, artificial transmutation, fission, or fusion) described.

Type of Nuclear Reaction	
1.) $^{42}_{19}\text{K} \rightarrow ^{0}_{-1}\text{e} + \text{_____}$	1.)
2.) $^{239}_{94}\text{Pu} \rightarrow ^{4}_{2}\text{He} + \text{_____}$	2.)
3.) $^{235}_{92}\text{U} \rightarrow \text{_____} + ^{231}_{90}\text{Th}$	3.)
4.) $^{2}_{1}\text{H} + ^{3}_{1}\text{H} \rightarrow \text{_____}$	4.)
5.) $^{6}_{3}\text{Li} + ^{1}_{0}\text{n} \rightarrow ^{4}_{2}\text{He} + \text{_____}$	5.)
6.) $^{27}_{13}\text{Al} + ^{4}_{2}\text{He} \rightarrow ^{30}_{15}\text{P} + \text{_____}$	6.)
7.) $^{9}_{4}\text{Be} + ^{1}_{1}\text{H} \rightarrow \text{_____} + ^{4}_{2}\text{He}$	7.)
8.) $^{37}_{19}\text{K} \rightarrow ^{0}_{-1}\text{e} + \text{_____}$	8.)
9.) $\text{_____} + ^{1}_{0}\text{n} \rightarrow ^{142}_{36}\text{Ba} + ^{91}_{36}\text{Kr} + 3 ^{1}_{0}\text{n}$	9.)
10.) $^{238}_{92}\text{U} + ^{4}_{2}\text{He} \rightarrow \text{_____} + ^{1}_{0}\text{n}$	10.)
11.) $^{14}_{6}\text{C} \rightarrow ^{14}_{7}\text{N} + \text{_____}$	11.)
12.) $^{187}_{75}\text{Re} + \text{_____} \rightarrow ^{188}_{75}\text{Re} + ^{1}_{1}\text{H}$	12.)
13.) $^{22}_{11}\text{Na} + \text{_____} \rightarrow ^{22}_{10}\text{Ne}$	13.) ---
14.) $^{218}_{84}\text{Po} \rightarrow \text{_____} + ^{4}_{2}\text{He}$	14.)
15.) $^{253}_{95}\text{Es} + ^{4}_{2}\text{He} \rightarrow ^{1}_{0}\text{n} + \text{_____}$	15.)

Write the balanced nuclear equations for the alpha decay of:

a) Plutonium-234

b) Strontium-90

Write the balanced nuclear equations for the alpha, beta and gamma decay of Radium-226

Chapter 252 Nuclear Transformations Answer Key

Stefan Luginbuehl, Peter Ganea

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