

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solvent and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is hydration in which the solvent is water. What is solvation?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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Chemistry: Matter and Change • Chapter 15

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CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Chapter 5 Study Guide Content Mastery Answer Key

Joan Ferrante-Wallace



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2 the problem of food choices 3 nutrient needs and nutrition surveys 4 energy balance overweight and underweight 5 food composition and safety and 6 measures for improving one's diet The following nutrients are examined in detail carbohydrates sugar starch fiber lipids proteins vitamins minerals and water Nutrient requirements throughout the life cycle i.e. pregnancy infancy childhood adolescence adulthood are also reviewed Supplementary topics include the role of doctors in nutrition the appropriateness of dietary guidelines nutrition and acne and nutrition and cancer Appendices list reliable and unreliable sources of nutrition information and present a series of self study exercises and accompanying forms for evaluating individual dietary and exercise habits aje *Im/Tb Cult Anthropology* Gary Ferraro,2005-04 *Basic Pharmacology for Nurses - E-Book* Michelle Willihnganz,Bruce D. Clayton,2014-05-01 Basic Pharmacology for Nurses is a modern classic Known for its impeccably accurate drug content and practical applications of the nursing process this trusted text consistently emphasizes health promotion through monitoring and patient education Introductory units ground your knowledge in the basic principles of pharmacology and medication administration while subsequent body system units prepare you to apply the nursing process to every major disorder UNIQUE Two tiered application of the nursing process provides the general principles of care along with specific nursing implications for each drug class or individual drug Meticulously accurate and updated drug content keeps the book in step with the latest FDA approvals withdrawals and changes in therapeutic uses UNIQUE Introduction to Cardiovascular Disease and Metabolic Syndrome sets the stage for an understanding of the progression of cardiovascular disease and the pharmacologic treatment of that progression UNIQUE Patient Teaching boxes equip you for health teaching related to medications LPN Threads design presents information in an easy to understand and visually engaging manner NEW Increased emphasis on medication safety prepares you for a clinical environment that is more focused than ever on reducing drug errors NEW Updated and added coverage keeps you up to date with the latest information on genetics pharmacogenomics and racial gender factors in drug actions **Mastery Study Guide Into Psychology** Tom Bond,Coon,1991-09 Study Guide for Berk, Child Development Patricia A. Jarvis,1991 Instructor Curriculum Guide for the Medical Assistant Mary E Kinn, CPS Cma-A,Kinn,Woods,1999-05 **Unlocking English Learners' Potential** Diane Staehr Fenner,Sydney Snyder,2017-05-16 Schools are not intentionally equitable places for English learners to achieve but they could be if the right system of support were put in place Diane Staehr Fenner and Sydney Snyder recommend just such a system Not only does it have significant potential for providing fuller access to the core curriculum it also provides a path for teachers to travel as they navigate the individual needs of students and support their learning journeys Douglas Fisher Coauthor of Visible Learning for Literacy A once in a generation text for assisting a new generation of students Content teachers and ESOL teachers take special note if you're looking for a single resource to help your English learners meet the same challenging content standards as their English proficient peers your search is complete Just dip into this toolbox of strategies examples templates and activities from EL authorities Diane Staehr Fenner

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