

## 2.2 Concept Review

### Section: Properties of Matter

1. Classify each of the following as a physical or chemical property of sulfur.

- \_\_\_\_\_ a. Its density is 2.97 g/cm<sup>3</sup>.
- \_\_\_\_\_ b. It reacts with hydrogen to form a gas.
- \_\_\_\_\_ c. It is a yellow solid.
- \_\_\_\_\_ d. Its melting point is 112°C.
- \_\_\_\_\_ e. It combines with oxygen.

2. Classify each of the following as a physical or chemical property of phosphorus.

- \_\_\_\_\_ a. It is a white, waxy solid.
- \_\_\_\_\_ b. It burns in air.
- \_\_\_\_\_ c. Its melting point is 44.1°C.
- \_\_\_\_\_ d. It has a density of 1.82 g/cm<sup>3</sup>.
- \_\_\_\_\_ e. Its boiling point is 280.3°C.

3. Explain why aluminum is a suitable material to use in making cans based on its physical and chemical properties.

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4. Calculate the mass of a sample of pure silver (density = 10.49 g/cm<sup>3</sup>) that has a volume of 12.99 cm<sup>3</sup>.

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5. Compute the density of an 820 g sample of pure silicon occupying a 350 cm<sup>3</sup> container.

## Concept Review Describing Matter Answer Key

**Allan Blackman, Daniel  
Southam, Gwendolyn Lawrie, Natalie  
Williamson, Christopher Thompson**

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