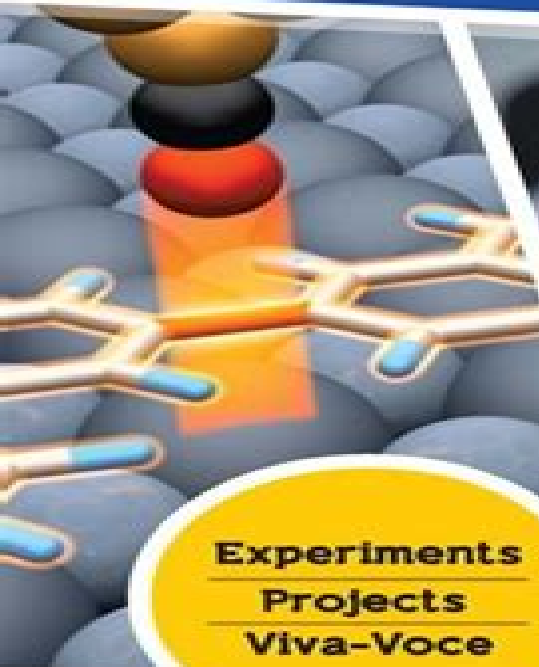


**LABORATORY MANUAL**

# Chemistry

## CBSE Class XII



**Experiments**  
**Projects**  
**Viva-Voce**



# Cbse Chemistry Lab Manual 2012 Class 12

**Emilie Sanchez**



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**Comprehensive Practical Chemistry XII** Dr. N . K. Verma, B. K. Vermani, Dr. Neera Verma, K . K. Rehani, 2011-11

**Lab Manual Chemistry Class XII -by Dr. K. N. Sharma, Dr. Subhash Chandra Rastogi, Er. Meera Goyal (SBPD Publications)** Dr. K. N. Sharma, Dr. Subhash Chandra Rastogi, Er. Meera Goyal, 2021-07-03 Highly Useful for Various Engineering and Medical Competitive Examinations **Chemistry 141 Lab Manual, Fall 2012**, 2012 **Chemistry Lab Manual** Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques

1 To cut a glass tube or glass rod

2 To bend the glass rod at an angle

3 To draw a glass jet from a glass tube

4 To bore a cork and fit a glass tube into it

Viva Voce

2 Characterisation and Purification of Chemical Substances

1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique

Viva Voce

2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method

Viva Voce

3 To prepare crystals of pure potash alum  $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$  from the given impure sample

4 To prepare the pure crystals of copper sulphate from the given crude sample

5 To prepare pure crystals of benzoic acid from a given impure sample

Viva Voce

3 Measurement of pH Values

1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper

2 To determine and compare the pH values of solutions of strong acid HCl and weak acid  $CH_3COOH$  of same concentration

3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper

4 To study the pH change by common ion  $CH_3COO^-$  ion in case of weak acid  $CH_3COOH$

5 To determine the change in pH value of weak base  $NH_4OH$  in presence of a common ion  $NH_4^+$

Viva Voce

4 Chemical Equilibrium

1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions

2 To study the shift in equilibrium between  $Co(H_2O)_6^{2+}$  and  $Cl^-$  ions by changing the concentrations of either of the ions

Viva Voce

5 Quantitative Analysis

1 To prepare M 10 oxalic acid solution by direct weighing method

2 To prepare M 10 solution of sodium carbonate by direct weighing method

3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid

4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

Viva Voce

6 Qualitative Analysis

Analysis of Anions

Analysis of Cations

Viva Voce

7 Detection of Elements in Organic Compounds

1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test

2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

Viva Voce

INVESTIGATORY PROJECTS

1 Checking of Bacterial Contamination in Water

1 To check the bacterial contamination in drinking water by testing sulphide ions

Viva Voce

2 Methods of Water Purification

1 To purify water from suspended impurities by using sedimentation

2 To purify water by boiling

3 To purify water by distillation method

4 To purify water by reverse osmosis technique

5 To purify water by GAC method

6 To purify water by bleach

treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids Viva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

**Chemistry 135 Lab Manual, Fall 2012** Bill Steel, 2012

**General Chemistry 117 Lab Manual, Fall 2012**, 2012

Principles of Chemistry I Lab Manual - SCI 123, Fall 2012

, 2012 Practical/Laboratory Manual Chemistry Class XII based on NCERT guidelines by Dr. S. C. Rastogi, Er. Meera Goyal

Dr. S. C. Rastogi, Er. Meera Goyal, 2020-06-22 A Surface Chemistry 1 To prepare colloidal solution sol of starch 2 To prepare a colloidal solution of egg albumin 3 To prepare colloidal solution of gum 4 To prepare colloidal solution of aluminium hydroxide  $\text{Al(OH)}_3$  5 To prepare colloidal solution of ferric hydroxide  $\text{Fe(OH)}_3$  6 To prepare colloidal solution of arsenious sulphide  $\text{As}_2\text{S}_3$  7 To purify a freshly prepared sol by dialysis 8 To compare the effectiveness of different common oils Castor oil cotton seed oil coconut oil kerosene oil mustard oil in forming emulsions Viva Voce B Chemical Kinetics 1 To study the effect of concentration on the rate of reaction between sodium thiosulphate and hydrochloric acid 2 To study the effect of temperature on the rate of reaction between sodium thiosulphate and hydrochloric acid 3 To study the rate of reaction of iodide ions with hydrogen peroxide at different concentrations of iodide ions 4 To study the rate of reaction between potassium iodate  $\text{KIO}_3$  and sodium sulphite  $\text{Na}_2\text{SO}_3$  using starch solution as indicator Viva Voce C Thermochemistry 1 Determine the enthalpy of dissolution of copper sulphate  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  in water at Room temperature 2 To determine the enthalpy of neutralization of the reaction between  $\text{HCl}$  and  $\text{NaOH}$  3 To determine enthalpy change during the interaction between acetone and chloroform Viva Voce D Electrochemistry 1 To study the variation of cell potential in  $\text{Zn}|\text{Zn}^{2+}||\text{Cu}^{2+}|\text{Cu}$  with change in concentration of electrolytes  $\text{CuSO}_4$  or  $\text{ZnSO}_4$  at room temperature Viva Voce E Chromatography 1 To separate the coloured components pigment present in the given extract of leaves and flowers by ascending paper chromatography and find their  $R_f$  values 2 To separate the coloured components present in the mixture of red and blue inks by ascending paper chromatography and find their  $R_f$  values 3 To separate  $\text{Co}^{2+}$  and  $\text{Ni}^{2+}$  ions present in the given mixture by using ascending paper chromatography and determine their  $R_f$  values Viva Voce F Preparation of Inorganic Compounds 1 Preparation of double salt of ferrous ammonium sulphate Mohr's salt from ferrous sulphate and ammonium sulphate 2 To prepare a pure sample of potash alum fitkari 3 Preparation of crystals of potassium ferric oxalate or potassium trioxalato ferrate III Viva Voce G Preparation of Organic Compounds 1 Preparation of iodoform from ethyl alcohol or acetone 2

Preparation of acetanilide in laboratory 3 Preparation of b Naphthol aniline dye 4 To prepare a pure sample of dibenzalacetone 5 To prepare a pure sample of p nitro acetanilide Viva Voce H Tests for the Functional Groups Present in Organic Compounds Viva Voce I Study of Carbohydrates Fats and Proteins 1 To study simple reactions of carbohydrate 2 To study simple reactions of fats 3 To study simple reactions of proteins 4 To investigate presence of carbohydrates fats and proteins in food stuffs Viva Voce J Volumetric Analysis 1 To prepare 250 ml of M 10 solution of oxalic acid 2 To prepare 250 ml of M 10 solution of ferrous ammonium sulphate 3 Prepare M 20 solution of oxalic acid with its help find out the molarity and strength of the given solution of potassium permanganate 4 Prepare M 20 solution of Mohr s salt using this solution determine the molarity and strength of potassium permanganate solution Viva Voce K Qualitative Analysis Viva Voce INVESTIGATORY PROJECTS 1 To study the presence of oxalate ions in guava fruit at different stages of ripening 2 To study the quantity of caseine present in different samples of milk 3 Preparation of soyabean milk and its comparison with natural milk with respect to curd formation effect of temperature etc 4 To study the effect of potassium bisulphite as food preservative at various concentrations 5 To study the digestion of starch by salivary amylase and the effect of pH and temperature on it 6 To study and compare the rate of fermentation of the following materials wheat flour gram flour potato juice and carrot juice 7 To extract essential oils present in saunf aniseed ajwain corum illaichi cardomom 8 To detect the presence of adulteration in fat oil and butter 9 To investigate the presence of NO<sub>2</sub> in brinjal      *Comprehensive Practical Chemistry (Abhilekhan) XII* Dr. N.K. Verma,Avinash Ojha,2008-05-01      **Comprehensive Practical Chemistry XI** Dr. N. K. Verma,B. K. Vermani,Dr. Neera Verma,2010-02      **Addison-Wesley Chemistry Laboratory Manual** Dennis D. Staley,1995 *Chemistry 132* Sharyl A. Majorski,2012-08-15      **Chem 111-112** Larry Alan Scheich,Tammy Janine Melton,New Mexico State University. Department of Chemistry and Biochemistry,2002      **Chemistry** Peter T. Wassell,2009-08-10      *CHEM 111 Lab Manual General Chemistry* Christofield,2020-08-12      **General Chemistry Lab Manual -** ,2014      *Chemistry 123 Laboratory Manual* Karen E. Eichstadt,1992-01-01      **Chemistry 123 Laboratory Manual** Karen Eichstadt,1997-03-01      General Chemistry Lab Manual for de Anza College Zvi Szafran,Ronald M. Pike,Mono M. Singh,Judith C. Foster,2012-10-23

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