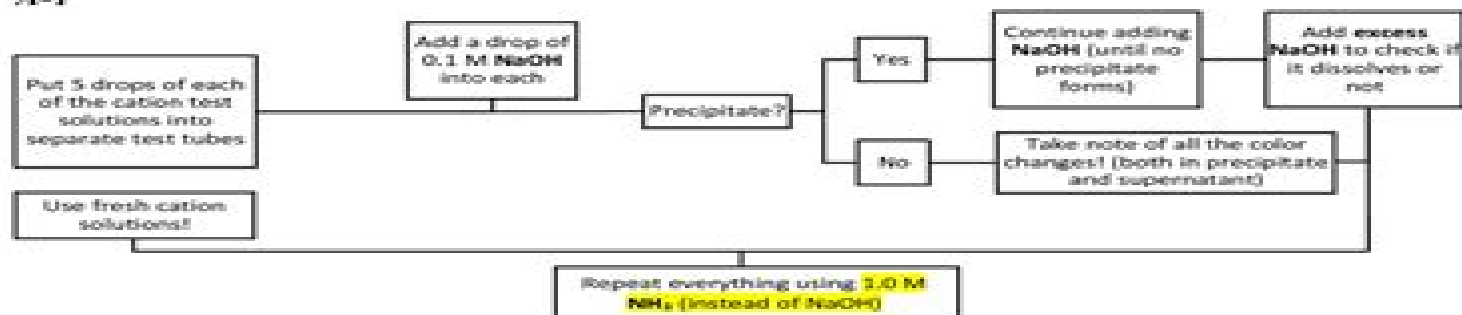





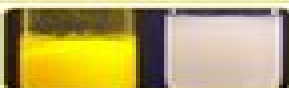
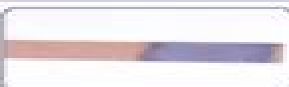
1. CATIONS – Ca^{2+} , Fe^{3+} , Cu^{2+} , Zn^{2+} , NH_4^+

Elimination Tests

A-1



Confirmatory Tests

| | |
|---|---|
|  | Cu^{2+} • Deep blue complex (upon mixing of solution with excess NH_3) |
|  | Fe^{3+} • Blood red complex (FeSCN^{2+}) (upon addition of 2 drops of 0.1 M KSCN) • Dark blue precipitate ($\text{KFe}^{III}\text{Fe}^{II}(\text{CN})_6$) (upon addition of 2 drops of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$) |
|  | Ca^{2+} • White precipitate (upon mixing with 2 drops of 0.1 M $(\text{NH}_4)_2\text{C}_2\text{O}_4$) |
|  | Zn^{2+} • Yellow precipitate ($\text{K}_2\text{ZnFe}(\text{CN})_6$) (upon addition of 2 drops of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$) • White precipitate ($\text{K}_2\text{Zn}(\text{C}_2\text{O}_4)_2$) (upon addition of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$) |
|  | NH_4^+ • Red litmus paper turns blue upon contact with vapor formed by mixing 30 drops of solution with 5 drops of 1.0 M NaOH |

Moisten 2 pieces of litmus paper (red and blue) with distilled water

Stick to the underside of watch glass

Immediately cover evaporating dish (containing the solution above) with the watch glass (litmus paper facing the solution)

Do NOT allow solution to touch the litmus paper!

Cations And Ions Lab Report

Rachel Sandford



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Prune Production Manual Richard P. Buchner, 2012-05-16 Written in easy to read non technical language this manual is the perfect field application guide Inside you ll find the professionalism expertise and science based answers you ve come to expect from the University of California with contributions from more than 40 Cooperative Extension professionals UC faculty USDA scientists and highly skilled prune industry experts Chapters include An industry overview A detailed description of prune biology Information on understanding soils varieties irrigation and fertilization Pest management techniques A lesson on harvest and postharvest management The breadth of expertise and knowledge contained in the 320 pages of this manual along with the more than 300 photos and 56 color illustrations make this one of the most comprehensive prune production manuals in the world

Pistachio Production Manual Louise Ferguson, David Haviland, 2016-12-02 The long awaited Pistachio Production Manual from the University of California is here The combined knowledge of 42 UC and industry experts and years of research and field trials are brought to fruition in this long awaited 321 page manual From an overview of the state of the industry to physiological disorders the 8 part manual covers everything you need to know Chapters cover topics including orchard design rootstocks and cultivars planting and training young trees weed insect mite and vertebrate management irrigation and salinity management disease management and physiological disorders including alternate bearing nut blanking and shell splitting Over 200 color photographs and 60 diagrams charts and tables illustrate key points The back cover includes a photographic guide to the developmental stages of the pistachio

Walnut Production Manual David E. Ramos, 1997-10-01

Laboratory Exercises for Preparatory Chemistry Kathy Dodds Tyner, 1994-06 Laboratory Exercises for Preparatory Chemistry is the perfect complement to a one semester preparatory chemistry laboratory course Tyner s manual emphasizes the application of chemistry and the principles of science to everyday life The labs are directly applicable to the real world and often contain supplemental assignments that illustrate an application

Annual Report - Agricultural Experiment Station Alabama Agricultural Experiment Station, 1906 *Annual Report of the Agricultural Experiment Station of the Alabama Polytechnic Institute* Alabama Polytechnic Institute. Agricultural Experiment Station, 1919

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Laboratory Exercises in Anatomy and Physiology with Cat Dissections Gerard J. Tortora, Robert B. Tallitsch, 1996 Following a body systems approach this laboratory manual is designed to be compatible with any introductory anatomy and physiology text It includes exercises which encourage microscopic examinations of cells observe chemical reactions perform dissections record data and analyze results

Experiment Station Record United States. Office of Experiment Stations, 1917 **Elements of General and Biological Chemistry, Laboratory Manual** John R. Holum, 1994-10-19 This updated edition explains recent advances in environmental studies and in the molecular basis of

life Suitable for students interested in the health care field as well as those who want to know how nature and human life work at the molecular level the book begins by providing readers with a solid background in formulas structures equations solutions and equilibria A number of topics are introduced early such as molarity and are discussed in more detail in later chapters Each chapter contains a summary as well as review exercises General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems *Chemistry in the Laboratory* James M. Postma, Julian L. Roberts, J. Leland Hollenberg, 2009-08-07 For nearly 40 years Chemistry in the Laboratory has been meeting the needs of teachers and students This new edition builds on that legacy while addressing cutting edge trends in the chemistry laboratory including forensic chemistry and environmental and green chemistry As always the new edition of Chemistry in the Laboratory offers precise easy to follow instructions helpful illustrations and an emphasis throughout on laboratory safety Again throughout a Consider This feature encourages users to expand the principles of the experiment into interesting applications open ended experiments or unexplored corners Most experiments in the manual can be completed in one lab session but some can be linked or extended for a multi lab project **Basic Laboratory Principles in General Chemistry** Fitzgerald B. Bramwell, 1990 **The Freeman Library of Laboratory Separates in Chemistry: 1083-1122**, 1970 **Chemistry of Life** Peter P. Berlow, 1982 U.S. Government Research & Development Reports, 1970 **Technical Note**, 1917 **Government Reports Announcements & Index**, 1988 Technical Information Pilot, 1949 **Qualitative Analysis and Ionic Equilibrium** George H. Schenk, 1990 A supplement for courses with a qualitative analysis component this lab manual contains explanations of the chemistry of metal ions and anions It includes pre lab exercises experiments and lab reports

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