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Lab Manual



Cbse 10th Ncert Science Lab Manual

Dr. N. K. Sharma, Bishan Lal



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A Basic Laboratory Techniques

- 1 To cut a glass tube or glass rod
- 2 To bend the glass rod at an angle
- 3 To draw a glass jet from a glass tube
- 4 To bore a cork and fit a glass tube into it

B Characterisation and Purification of Chemical Substances

- 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
- 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
- 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
- 4 To prepare the pure crystals of copper sulphate from the given crude sample
- 5 To prepare pure crystals of benzoic acid from a given impure sample

C Measurement of pH Values

- 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
- 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
- 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
- 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
- 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+

D Chemical Equilibrium

- 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
- 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions

E Quantitative Analysis

- 1 To prepare M 10 oxalic acid solution by direct weighing method
- 2 To prepare M 10 solution of sodium carbonate by direct weighing method
- 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
- 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

F Qualitative Analysis

- 1 Analysis of Anions
- 2 Analysis of Cations

G Detection of Elements in Organic Compounds

- 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
- 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

INVESTIGATORY PROJECTS

A Checking of Bacterial Contamination in Water

- 1 To check the bacterial contamination in drinking water by testing sulphide ions

B Methods of Water Purification

- 1 To purify water from suspended impurities by using sedimentation
- 2 To purify water by boiling
- 3 To purify water by distillation method
- 4 To purify water by reverse osmosis technique
- 5 To purify water by GAC method
- 6 To purify water by bleach treatment
- 7 To purify water by oxidising agent
- 8 To purify water by ozone treatment method

C Water Analysis

- 1 To test the hardness of different water samples

D Foaming Capacity of Various Soaps

- 1 To compare the foaming capacity of different washing soaps
- 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap

E Tea Analysis

- 1 To study the acidity of different samples of tea leaves tea by using pH paper

F Analysis of Fruits and Vegetable Juices

- 1 To analyse the fruit and vegetable juices for the constituent present in them

G Rate of Evaporation

- 1 To study the rate of evaporation of different

liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table

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scientific concepts taught in their textbooks but also to ignite a scientific quest in their young inquisitive minds
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Table of Contents Cbse 10th Ncert Science Lab Manual

1. Understanding the eBook Cbse 10th Ncert Science Lab Manual
 - The Rise of Digital Reading Cbse 10th Ncert Science Lab Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Cbse 10th Ncert Science Lab Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Cbse 10th Ncert Science Lab Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Cbse 10th Ncert Science Lab Manual
 - Personalized Recommendations
 - Cbse 10th Ncert Science Lab Manual User Reviews and Ratings
 - Cbse 10th Ncert Science Lab Manual and Bestseller Lists
5. Accessing Cbse 10th Ncert Science Lab Manual Free and Paid eBooks

- Cbse 10th Ncert Science Lab Manual Public Domain eBooks
 - Cbse 10th Ncert Science Lab Manual eBook Subscription Services
 - Cbse 10th Ncert Science Lab Manual Budget-Friendly Options
6. Navigating Cbse 10th Ncert Science Lab Manual eBook Formats
- ePub, PDF, MOBI, and More
 - Cbse 10th Ncert Science Lab Manual Compatibility with Devices
 - Cbse 10th Ncert Science Lab Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Cbse 10th Ncert Science Lab Manual
 - Highlighting and Note-Taking Cbse 10th Ncert Science Lab Manual
 - Interactive Elements Cbse 10th Ncert Science Lab Manual
8. Staying Engaged with Cbse 10th Ncert Science Lab Manual
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Cbse 10th Ncert Science Lab Manual
9. Balancing eBooks and Physical Books Cbse 10th Ncert Science Lab Manual
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Cbse 10th Ncert Science Lab Manual
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Cbse 10th Ncert Science Lab Manual
- Setting Reading Goals Cbse 10th Ncert Science Lab Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Cbse 10th Ncert Science Lab Manual
- Fact-Checking eBook Content of Cbse 10th Ncert Science Lab Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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