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# SCIENCE

## Lab Manual



# Cbse 10th Ncert Science Lab Manual

**Dr. N. K. Sharma, Bishan Lal**

## **Cbse 10th Ncert Science Lab Manual:**

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**Comprehensive Practical Science IX** S.P. Saxena,Dheeraj Saxena,2010     *Lab Manual Science Class 10* Neena Sinha, R.Rangarajan, Rajesh Kumar, These Lab Manuals provide complete information on all the experiments listed in the latest CBSE syllabus The various objectives materials required procedures inferences etc have been given in a step by step manner Carefully framed MCQs and short answers type questions given at the end of the experiments help the students prepare for viva voce

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*Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal* Dr. S. C. Rastogi , Er. Meera Goyal, 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards

**A Basic Laboratory Techniques**

- 1 To cut a glass tube or glass rod
- 2 To bend the glass rod at an angle
- 3 To draw a glass jet from a glass tube
- 4 To bore a cork and fit a glass tube into it

**B Characterisation and Purification of Chemical Substances**

- 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
- 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
- 3 To prepare crystals of pure potash alum  $K_2SO_4 Al_2 SO_4 3 24H_2O$  from the given impure sample
- 4 To prepare the pure crystals of copper sulphate from the given crude sample
- 5 To prepare pure crystals of benzoic acid from a given impure sample

**C Measurement of pH Values**

- 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
- 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid  $CH_3COOH$  of same concentration
- 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
- 4 To study the pH change by common ion  $CH_3COO$  ion in case of weak acid  $CH_3COOH$
- 5 To determine the change in pH value of weak base  $NH_4OH$  in presence of a common ion  $NH_4$

**D Chemical Equilibrium**

- 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
- 2 To study the shift in equilibrium between  $Co^{2+}$  and  $Cl^-$  ions by changing the concentrations of either of the ions

**E Quantitative Analysis**

- 1 To prepare M 10 oxalic acid solution by direct weighing method
- 2 To prepare M 10 solution of sodium carbonate by direct weighing method
- 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
- 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

**F Qualitative Analysis**

- 1 Analysis of Anions
- 2 Analysis of Cations

**G Detection of Elements in Organic Compounds**

- 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
- 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

**INVESTIGATORY PROJECTS**

- A Checking of Bacterial Contamination in Water
- 1 To check the bacterial contamination in drinking water by testing sulphide ions
- B Methods of Water Purification

  - 1 To purify water from suspended impurities by using sedimentation
  - 2 To purify water by boiling
  - 3 To purify water by distillation method
  - 4 To purify water by reverse osmosis technique
  - 5 To purify water by GAC method
  - 6 To purify water by bleach treatment
  - 7 To purify water by oxidising agent
  - 8 To purify water by ozone treatment method

**C Water Analysis**

- 1 To test the hardness of different water samples

**D Foaming Capacity of Various Soaps**

- 1 To compare the foaming capacity of different washing soaps
- 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap

**E Tea Analysis**

- 1 To study the acidity of different samples of tea leaves tea by using pH paper

**F Analysis of Fruits and Vegetable Juices**

- 1 To analyse the fruit and vegetable juices for the constituent present in them

**G Rate of Evaporation**

- 1 To study the rate of evaporation of different

liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table

**Academic Practical Science X** Dr. N. K. Sharma, Bishan Lal, 2011-12    **Annual Report** India. Department of School Education & Literacy, 2007    *Comprehensive Practical Science X* S. P. Saxena, Dheeraj Saxena, 2011-12    *Science Lab Manual Stage 10* Dr. Kamalkant, Dr. Bharati Khanna, Dr. Manmohan Singh, 2011-01-01    Practical/Laboratory Manual Science Class X based on NCERT guidelines by Dr. J. P. Goel, Dr. S. C. Rastogi, Dr. Sunita Bhagia & Er. Meera Goyal Dr. J. P. Goel, Dr. S. C. Rastogi, Dr. Sunita Bhagia, Er. Meera Goyal, 2020-06-26

Physics 1 To determine the focal length of concave mirror 2 To find the focal length of convex lens by two pin method 3 To find the image distance for varying object distances in case of a convex lens and drawing corresponding ray diagrams to show the nature of image formed 4 To trace the path of the rays of light through a glass prism 5 To trace the path of a ray of light passing through a rectangular glass slab for difference angles of incidence 6 To study the dependence of potential difference V across a resistor on the current I passing through it and determine its resistance Also plotting a graph between V and I 7 To determine the equivalent resistance of two resistors when connected in series and parallel

Chemistry 8 To find the pH of the following samples by using pH paper universal indicator 9 To studying the properties of a base dil NaOH Solution and Acid HCl by their reaction with a Litmus solution Blue Red b Zinc metal c Solid sodium carbonate 10 To perform and observe the following reactions and to classify them into a Combination reaction b Decomposition reaction c Displacement reaction d Double displacement reaction i Action of water on quick lime ii Action of heat on ferrous sulphate crystals iii Iron nails kept in copper sulphate solution iv Reaction between sodium sulphate and barium chloride solutions 11 To observe the action of Zn Fe Cu and Al on the following salt solutions a ZnSO<sub>4</sub> aq b FeSO<sub>4</sub> aq c CuSO<sub>4</sub> aq d Al<sub>2</sub>SO<sub>4</sub> 3 aq Based on the above result to arrange Zn Fe Cu and Al metals in the decreasing order of reactivity 12 To study the following properties of acetic acid ethanoic acid i Odour ii Solubility in water iii Effect on litmus iv Reaction with sodium hydrogen carbonate 13 To study the comparative cleaning capacity of a sample of soap in soft and hard water

Biology 14 To study stomata by preparing a temporary mount of a leaf peel 15 To show experimentally that carbon dioxide CO<sub>2</sub> is given out during aerobic respiration 16 To study A Binary fission in Amoeba and B Budding in yeast with the help of prepared slides 17 To identify the different parts of an embryo of a dicot seed pea gram or red kidney beans

**Core Science Lab Manual with Practical Skills for Class X** V. K. Sally, Chhaya Srivastava, Goyal Brothers Prakashan, 2019-01-17

**Goyal Brothers Prakashan**    **Laboratory Manual for Science 10** A. K. Raj, Laboratory Manual for Science is a series of five books for classes 6 to 10 These are complimentary to the Science textbooks of the respective classes The manuals cover a wide range of age appropriate experiments that give hands on experience to the students The experiments help students verify scientific truths and principles and at the same time expose them to the basic tools and techniques used in scientific investigations Our manuals aim not only to help students better comprehend the

scientific concepts taught in their textbooks but also to ignite a scientific quest in their young inquisitive minds [Complete Science Laboratory Manual CBSE For Class 9 Education Experts,2019-10](#)

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## **Table of Contents Cbse 10th Ncert Science Lab Manual**

1. Understanding the eBook Cbse 10th Ncert Science Lab Manual
  - The Rise of Digital Reading Cbse 10th Ncert Science Lab Manual
  - Advantages of eBooks Over Traditional Books
2. Identifying Cbse 10th Ncert Science Lab Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Cbse 10th Ncert Science Lab Manual
  - User-Friendly Interface
4. Exploring eBook Recommendations from Cbse 10th Ncert Science Lab Manual
  - Personalized Recommendations
  - Cbse 10th Ncert Science Lab Manual User Reviews and Ratings
  - Cbse 10th Ncert Science Lab Manual and Bestseller Lists
5. Accessing Cbse 10th Ncert Science Lab Manual Free and Paid eBooks

- Cbse 10th Ncert Science Lab Manual Public Domain eBooks
- Cbse 10th Ncert Science Lab Manual eBook Subscription Services
- Cbse 10th Ncert Science Lab Manual Budget-Friendly Options

6. Navigating Cbse 10th Ncert Science Lab Manual eBook Formats

- ePub, PDF, MOBI, and More
- Cbse 10th Ncert Science Lab Manual Compatibility with Devices
- Cbse 10th Ncert Science Lab Manual Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Cbse 10th Ncert Science Lab Manual
- Highlighting and Note-Taking Cbse 10th Ncert Science Lab Manual
- Interactive Elements Cbse 10th Ncert Science Lab Manual

8. Staying Engaged with Cbse 10th Ncert Science Lab Manual

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Cbse 10th Ncert Science Lab Manual

9. Balancing eBooks and Physical Books Cbse 10th Ncert Science Lab Manual

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Cbse 10th Ncert Science Lab Manual

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Cbse 10th Ncert Science Lab Manual

- Setting Reading Goals Cbse 10th Ncert Science Lab Manual
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Cbse 10th Ncert Science Lab Manual

- Fact-Checking eBook Content of Cbse 10th Ncert Science Lab Manual
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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