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Mathematics Enrichment Lab Activities 10 S. N. Sharma, R. S. Dhauni, Goyal Brothers Prakashan, 2020-04-01 Goyal Brothers Prakashan *Annual Report* India. Department of School Education & Literacy, 2007 Maths Mate □ 8 NEW Madhubun, 1 It is a series of eight textbooks for Classes 1 to 8 that conforms to the vision of National Curriculum Framework and is written in accordance with the latest syllabus of the CBSE 2 Learning Objectives Lists well what a learner will know and be able to do after studying the chapter 3 Let's Recall Refreshes the concepts learnt in the form of a revision exercise to brush up the concepts taught in previous chapters or grades 4 Let's Begin Introduction to the chapter 5 My Notes Tips to help the learner remember the important points formulae taught in the chapter 6 Let's Try Simple straight forward questions for quick practice while studying any topic based on the first two levels of Bloom's Taxonomy Knowledge and Understanding 7 Error Alarm Common mistakes which learners commit often along with the correct way of doing the same 8 Know More Additional information for the learners relating to the concepts learnt in the chapter 9 Maths in My Life includes questions relating Maths to daily life and which can help relate the topic with the environment life around us 10 Tricky Maths Challenge questions to help the learners build thinking skills and reasoning skills by solving tricky questions 11 Project Work Projects which can help learners connect Math with our daily life or that take the concepts learnt to a new level 12 Concept Map Summary points to list the important concepts learnt in the chapter in a crisp form 13 Test Zone Revision exercise of the concepts learnt in the chapter This includes both objective and subjective type of questions 14 Mental Maths Maths problems

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Activities for Teachers of Secondary Mathematics Gerald Kulm, 1976 *Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal* Dr. S. C. Rastogi, Er. Meera Goyal, 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards A Basic Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it B Characterisation and Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample C Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ D Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions E Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution F Qualitative Analysis 1 Analysis of Anions 2 Analysis of Cations G Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test INVESTIGATORY PROJECTS A Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions B Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by

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