



67 Productivity Hacks

To Make You More Efficient

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Productivity Guide Fan Favorite

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Fan Fiction Genres Julia Elena Goldmann, 2022-09-01 What if James T Kirk and Spock had a baby left the Enterprise and moved to New Vulcan to live happily ever after Fan fiction plots like this are a strong testament of fans endless creativity Not only do the authors invent their own storylines but they have developed a generic definition of content across fandoms according to the relationship present in the text Classification is therefore profoundly related to gender and sexuality Julia Elena Goldmann examines these generic structures and formulaic patterns comparatively in Star Trek and Supernatural fan fiction She also focuses on the interplay of the concepts of gender sexuality relationships and depictions of family in these texts

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web using algebra the moles was founded to be 0.00352 moles of butane in order to find the molarity the mass of the butane should be divided by the moles after further calculations the molarity of the butane gas was calculated to be 56.81 grams moles this lab is done with tons of sources of error

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web this data shows that there was a gain the mass of the lighter which shows that there was a certain amount of butane 0.38 grams 0.02 grams gas that was successfully collected during the gas collection there is a mild influence of barometric pressure during both trials as shown above in the raw data

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web 11 use avogadro's law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22.4 l at stp $V_1/n_1 = V_2/n_2$ $0.0896 \text{ l} / n_2 = 22.4 \text{ l} / 1 \text{ mole}$ $n_2 = 0.0040 \text{ moles}$ 12 experimental molar mass g mole of butane 0.23 g / 0.0040 moles 58 g mole 13 the molecular formula of butane is C_4H_{10}

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web nov 3 2020 in this experiment the moles and mass will be required to determine the molar mass of butane in a butane lighter experimental to begin the experiment the mass of a butane lighter was measured before using the butane gas to deplete the water to the 80ml mark on a graduated cylinder

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web the pressure of the gas the pressure of the gas was calculated using dalton s law of partial pressures by subtracting the pressure of the water vapor from the total pressure which in this case was the atmospheric pressure 760mm hg the pressure of the water vapor was found by using the temperature of the water

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web use the combined gas law to determine the volume in l of butane at stp use avogadro s law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22 l at stp experimental molar mass g mole of butane the molecular formula of butane is c 4 h 10

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web although the most convenient gas for this is butane other gases may be available in similar small cylinders the collection of a gas sample and the weighing of the gas cylinder before and after this should take about 5 10 minutes as a demonstration

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
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