

Chapter 6: Structure of Plasmodesm and Kupffer bodies

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Plasmodesm Production

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Most students agree that the following model explains:

- (i) Plasmodesm production is guided by biological membranes.
Biological membranes are composed of phospholipid bilayers.

Biological Membrane Model

- (ii) Membrane movement and transport occur in biological membranes.
Membrane transport requires a carrier protein (or channel protein).

Cell Wall and Cell Wall Model

- (iii) Cell walling forces are exerted by proteins and the proteins depend on the underlying membranes.
Protein molecules undergo passive diffusion, protein active transport, or membrane-associated protein transport.
Protein transport is a protein carrier protein or channel protein.

Cell Wall and Cell Wall Model

Most students agree that the following statement:

- (i) The cell wall is a membrane system with proteins, carbohydrates, and lipids.

Plant Membrane Model/Plant Cell Wall Model

- (ii) Lipoproteins, carbohydrate proteins, and glycoproteins are found in the cell wall and have similar structures.
Carbohydrates, proteins, and lipids.

Cell Wall and Cell Wall Model

- (iii) Cell walling is a protein-carbohydrate system in which the carbohydrate is attached to a protein molecule. If this were true, proteins would be found in the cell wall and carbohydrates would be found in the cell wall.
Proteins, carbohydrates, and lipids in the cell wall.

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Morris Hein, Susan Arena

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Michael S. Leonard, 2025-02-25 Calculating theoretical and percent yield is a fundamental skill for the laboratory. This book primarily targets Organic Chemistry Laboratory courses at the high school or college and university level as a supplemental resource to help students master this skill. It begins with simple examples from everyday life demonstrates the importance of balancing the equation addresses the role of the mole in these computations discusses different types of liquids considers the role of significant figures and culminates with the planning of syntheses. There are suggestions for further reading as well as practice problems and questions to ensure mastery. Begins with examples from everyday life that enable students to understand the concepts of theoretical and percent yield before applying those concepts to the laboratory. Addresses the necessity of balancing the reaction equation the centrality of the mole in these calculations and the role of significant figures in reporting the answer. Explains how to approach the calculations when using neat liquids or solutions. The culmination of this text is the use of the same thought processes to plan the amounts of reactants needed for syntheses of desired quantities of product. All of the problems in the book include detailed solutions with accompanying text to explain the answers and ancillaries also include suggestions for further reading

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