

Intro. to Chemical Reactions

Electrolyte Reactions

1. Write down formulas for the following compounds:

a) Barium hydroxide $\text{Ba}(\text{OH})_2$

b) Sodium phosphate Na_3PO_4

c) Magnesium phosphate $\text{Mg}_3(\text{PO}_4)_2$

d) Calcium iodide CaI_2

2. Write the formulas for the following acids:

a) HClO₄ Hydrogen perchlorate

b) H₂SO₄ Hydrogen sulfate

c) H₃PO₄ Hydrogen phosphate

d) H₂SiO₃ Hydrogen silicic acid

Classification of Reactions: Physical-chemical Reactions

There is often a chemical reaction between two substances:



There are reactions that you can write as chemical equations alone:

Chemical reactions result in the production and yield of products and reactants.

There is often an interrelationship between the reactants and the products but this is not always a direct connection.

Chemical Reactions: Definitions

1. Given the following reaction, write the equation that goes along with it:



Chemical reactions involve reactants and products and reactions are reversible.

Chemquest 22 Answer Key

David Kirk

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