

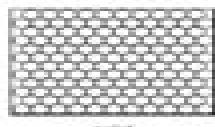
> Workbook answers

Chapter 1

Exercise 1.1

1. A solid has a fixed mass and volume / shape.
A liquid has a fixed mass but its shape changes to that of the container in which it is placed.
A gas has no fixed shape or volume. A gas completely fills the container that it is in.

17



10 of 10



10



10

3 a B
b C
c A
d D

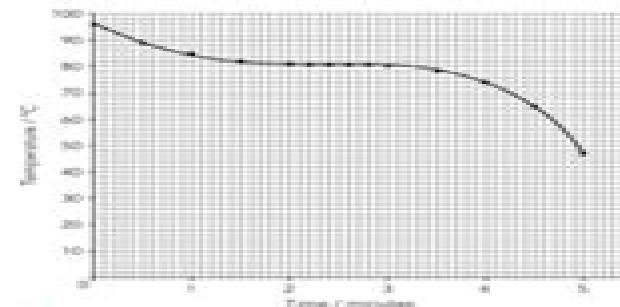
4 A freezing (solidification)
B melting
C condensation
D evaporation or boiling

5 a radon
b radon and nitrogen
c nitrogen
d cobalt

6 The sample of ethanoic acid is impure.
The presence of impurities raises the
boiling point of a substance.

Exercise 1.2

10



- 7 The student would need to use an oil bath (in place of the water bath) so that the higher temperature could be reached.
- 8 The substance is freezing (solidifying) / turning from liquid to solid.
- 9 The temperature stays constant because energy is being released as the substance solidifies / the molecules are giving out heat as they stop moving from place to place and become organised in a structured lattice arrangement / in the solid the molecules can only vibrate about fixed points / the heat released by the formation of new interactions (forces) between the particles keeps the temperature constant until all the substance is solid.

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1. Understanding the eBook Chemistry Workbook Answer Guide
 - The Rise of Digital Reading Chemistry Workbook Answer Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Workbook Answer Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Workbook Answer Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Workbook Answer Guide
 - Personalized Recommendations
 - Chemistry Workbook Answer Guide User Reviews and Ratings
 - Chemistry Workbook Answer Guide and Bestseller Lists
5. Accessing Chemistry Workbook Answer Guide Free and Paid eBooks
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8. Staying Engaged with Chemistry Workbook Answer Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry Workbook Answer Guide

9. Balancing eBooks and Physical Books Chemistry Workbook Answer Guide

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Chemistry Workbook Answer Guide

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Chemistry Workbook Answer Guide

- Setting Reading Goals Chemistry Workbook Answer Guide
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry Workbook Answer Guide

- Fact-Checking eBook Content of Chemistry Workbook Answer Guide
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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