

CHEMISTRY CHEAT SHEET GAS LAWS

GAS LAWS CHEAT SHEET

Gas Laws are mathematical expressions of the behavior of gases. They are derived from the kinetic molecular theory of gases, which states that all matter is composed of atoms or molecules in constant motion. The behavior of gases can be described by four fundamental laws: Boyle's Law, Charles' Law, Avogadro's Law, and the Ideal Gas Law.

Law	Equation
Boyle's Law	$P_1V_1 = P_2V_2$
Charles' Law	$\frac{V_1}{T_1} = \frac{V_2}{T_2}$
Avogadro's Law	$\frac{V_1}{N_1} = \frac{V_2}{N_2}$
Ideal Gas Law	$PV = nRT$

Boyle's Law: The pressure of a gas is inversely proportional to its volume. As the volume of a gas decreases, its pressure increases, and vice versa. This law is also known as the law of inverse proportions.



Chemistry Study Guide Gas Laws

Darrell D. Ebbing, Susan M. Schelble

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Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change *Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today* Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application. **Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl** Paul B. Kelter, 1993 *Chemistry Super Review - 2nd Ed.* Editors of

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