

## General Chemistry – Unit 8 Review

1. Vocabulary -

- a. mole
- b. molar mass
- c. Avogadro's number
- d. empirical formula
- e. molecular formula

2. Find the molar mass of the following:

a.  $\text{KNO}_3$  \_\_\_\_\_

d. oxygen gas \_\_\_\_\_

b.  $(\text{NH}_4)_2\text{CO}_3$  \_\_\_\_\_

e. calcium phosphate \_\_\_\_\_

c.  $\text{Ag}_2\text{CrO}_4$  \_\_\_\_\_

f. lead (II) nitrate \_\_\_\_\_

3. Consider the masses of various hardware below.

Type	Mass (g)	Relative mass
Washer	1.74	
Hex nut	3.16	
Anchor		3.00
Bolt	7.64	

a. Do the calculations necessary to complete the table.

b. Explain the connection between these calculations and the atomic masses in the Periodic Table.

# Chemistry Unit 8 Review Answers

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