

Lab Data

Volume of DI water (mL)

49.0

51.0

49.8

Verify your calculation

Moles of calcium chloride (mol)	0.0100	0.0196	0.0242
Mass of calorimeter + solution (g)	67.16	70.55	69.68
Mass of calorimeter + solution + ice (g)	112.82	115.83	116.45
Freezing-point of solution (°C)	-0.88	-1.53	-1.69
Mass of calorimeter + ice after decanting (g)	43.92	41.32	33.65
Mass of ice added (g)	45.66	45.28	46.77
Mass of ice remaining after decanting (g)	26.64	23.86	16.24
Mass of melted ice (g)	19.02	21.42	30.53
Mass of water in final solution (g)	67.79	72.34	80.11
Mass of water in final solution (kg)	0.06779	0.07234	0.08011
Molality of solution (m)	0.148	0.271	0.302
Freezing-point depression (°C)	0.78	1.43	1.59

van't Hoff factor



Colligative Properties Freezing Point Depression Lab Answers

**Willis Conway Pierce, Robert Nelson
Smith**

Colligative Properties Freezing Point Depression Lab Answers:

Oswaal CBSE Competency Focused Questions (with MCQs & Case/Source Based Integrated Questions) | Chapter-wise | Class 12 | Science Stream (PCM) | Physics | Chemistry | Mathematics | For 2025 Exam Oswaal Editorial Board, 2024-08-06

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Laboratory Experiments for Brown and LeMay, Chemistry, the Central Science John Henry Nelson,Kenneth C. Kemp,1977 Chemistry in the Laboratory James M. Postma,Julian L. Robert,J. Leland Hollenberg,2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

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Chemistry

McGraw-Hill Staff,2001-07 **General Chemistry Workbook** Willis Conway Pierce,Robert Nelson Smith,1965 **The Laboratory Handbook of Materials, Equipment, and Technique** Gary S. Coyne,1992 Using step by step procedures this book details the preparation storage cleaning care and maintenance for chemistry equipment Common difficulties are covered and techniques and procedures that make work in the laboratory more efficient productive and safe are suggested

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Mary Millhollen, Richard H. Langley, 2022-08-01 AP Teachers 1 Choice Ready to succeed in your AP course and ace

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why things are done in the lab and includes helpful guidelines and step by step procedures for each topic discussed back

cover

Proceedings Malcolm Correll, Arnold A. Strassenburg, 1965

Basic Concepts of Chemistry

Alan Sherman, Sharon Sherman, Leonard Russikoff, 1984

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Research Projects in Dry

Eye Syndrome

Horst Brewitt, 2010 Dry eye is one of the most common disorders encountered in ophthalmological practice

Its symptoms cause considerable discomfort and substantially reduce the patient's quality of life As it is a complex and

multifactorial condition research investigating dry eye is a matter of great interest all over the world In this book the wide

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scientists and clinical ophthalmologists to quickly bring themselves up to date with cutting edge research in this field This

overview of research into the pathogenesis and clinical treatment of dry eye is a must for all ophthalmologists dealing with this syndrome and looking to increase their knowledge base Laboratory Inquiry in Chemistry Richard C. Bauer, Richard Bauer, James P. Birk, Douglas J. Sawyer, 2005 LABORATORY INQUIRY IN CHEMISTRY Second Edition provides a unique set of guided inquiry investigations that focus on constructing knowledge about the conceptual basis of laboratory techniques instead of simply learning techniques. By focusing on developing skills for designing experiments, solving problems, thinking critically and selecting and applying appropriate techniques, the authors expose students to a realistic laboratory experience typical of the practicing chemist. The Second Edition features six new experiments and is accompanied by a revised and updated Instructor's Manual available online. This new edition continues the proven three phase learning cycle exploration of chemical behaviors within the context of the problems posed: concept invention, the use of data and observations to construct accepted scientific knowledge about the concepts explored in the laboratory investigation and concept application where students apply their conceptual understanding of the investigation at hand by modifying or extending the experiments and write a report that emphasizes conceptual relevance. These college and honors level inquiry based experiments correlate well with the recommended experiments outlined by the Advanced Placement Chemistry Development Committee.

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