

Lab Data

Volume of DI water (mL)

49.0

51.0

49.8

Verify your calculation

Moles of calcium chloride (mol)	0.0100	0.0196	0.0242
Mass of calorimeter + solution (g)	67.16	70.55	69.68
Mass of calorimeter + solution + ice (g)	112.82	115.83	116.45
Freezing-point of solution (°C)	-0.88	-1.53	-1.69
Mass of calorimeter + ice after decanting (g)	43.92	41.32	33.65
Mass of ice added (g)	45.66	45.28	46.77
Mass of ice remaining after decanting (g)	26.64	23.86	16.24
Mass of melted ice (g)	19.02	21.42	30.53
Mass of water in final solution (g)	67.79	72.34	80.11
Mass of water in final solution (kg)	0.06779	0.07234	0.08011
Molality of solution (m)	0.148	0.271	0.302
Freezing-point depression (°C)	0.78	1.43	1.59

van't Hoff factor



Colligative Properties Virtual Lab Answers Key

Michael Brown

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Observing Colligative Properties ,2009-01-01 Inquiries in Science Chemistry Series Observing Colligative Properties Teacher's Guide

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