

Chem 11, Spring 2012 Practice Test 2

1) In ionic compounds, _____ lose their valence electrons to form positively charged _____.

- A) metals, anions
- B) nonmetals, cations
- C) metals, polyatomic ions
- D) nonmetals, anions
- E) metals, cations

2) An anion always:

- A) has a positive charge.
- B) contains a group of two or more atoms with a positive charge.
- C) contains a metal and a nonmetal.
- D) forms covalent bonds.
- E) has a negative charge.

3) What is the ionic charge of an ion with 18 protons and 15 electrons?

- A) $1+$
- B) $2+$
- C) $3+$
- D) $1-$
- E) $3-$

4) The correct formula for a compound formed from the elements Al and O is:

- A) AlO .
- B) Al_2O .
- C) Al_2O_3 .
- D) AlO_3 .
- E) Al_2O_5 .

5) A(n) _____ is the smallest neutral unit of two or more atoms held together by a covalent bond.

- A) ionic compound
- B) nucleus
- C) molecule
- D) formula
- E) unit

Chem 11 Spring 2012 Practice Test

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