

Finals Grinding Chapter 1

1.1 Chemistry in Context

Scientific Methods

- Negative Result

Experiment/Observation → Result/Conclusion → hypothesis →
Communication More Experiments → Theory → Positive Results
Theory →
Theories & Laws →

1.2 Phases in Matter Classification

States of Matter Solid - fixed shape and volume
Liquid - shape of container, fixed volume
gas - expands to fill container
plasma - gaseous state w/ charged particles w/ unique properties

Mass & Weight - amount of matter
Weight - force of gravity on object
Conservation of Mass - mass doesn't change due to chemical or physical reactions
Pure Substance - constant composition

Elements same type of atom simplest version.
Compounds can be decomposed to other compounds of elements.
Homogeneous mixture uniform composition.
Heterogeneous mixture physical mix. at different points.

miscible, forms a homogeneous mixture when mixed.

1.3 Physical & Chemical Properties

Ex/Intensive Properties Extensive depend on size Intensive independent of size

Atomic number = протоны
Periods = ярусы
Groups =族

Chiroptera - groups 16 (o, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16) Groups - down.

Strong Acids: HClO_4 hydrochloric acid, HNO_3 nitric acid, HBr hydrobromic acid, HCl hydrochloric acid, H_2SO_4 sulfuric acid, H_3PO_4 phosphoric acid, HClO_3 chloric acid, H_2SO_3 sulfurous acid, H_2SeO_3 selenous acid, H_2TeO_3 telluric acid, HClO_2 chlorous acid, H_2SeO_4 selenic acid, H_2TeO_4 telluric acid.

1.4 Measurements

UNIT - Standard of Comparison
magnitude - size of number
indication of magnitude - size of number

indication of uncertainty - one uncertain digit
B100 100 2000 100 100 100 100 100 100 100

Pico 10^{-12} nano 10^{-9} micro 10^{-6} milli 10^{-3} centi 10^{-2} deci 10^{-1}
 Kilo 10^3 mega 10^6

**16. Denitro-
mesopholamine (invasive property)**

1.5 Uncertainty, Accuracy, Precision

measurement devices, user introduce uncertainty into the flow profile of particle 12.

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James E. Brady,John R. Holm

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