

Intermolecular

forces are weakly and long-range, all these molecules are grouped as forces of the nature of different intermolecular forces are intermolecular. E.g., H_2 , O_2 , N_2 .

Most of the intermolecular forces are caused by weak attractive forces. The following forces are responsible for the intermolecular forces: (i) London dispersion forces, (ii) dipole-dipole forces, (iii) hydrogen bonding.

Hydrogen Bonding

1. It occurs when every hydrogen atom is bonded with a highly electronegative atom like F , O or N .
2. It is present in all the molecules having electronegative atoms like F , O or N .
3. It is present in all the molecules having electronegative atoms like F , O or N .
4. It is present in all the molecules having electronegative atoms like F , O or N .
5. It is present in all the molecules having electronegative atoms like F , O or N .
6. It is present in all the molecules having electronegative atoms like F , O or N .
7. It is present in all the molecules having electronegative atoms like F , O or N .
8. It is present in all the molecules having electronegative atoms like F , O or N .
9. It is present in all the molecules having electronegative atoms like F , O or N .
10. It is present in all the molecules having electronegative atoms like F , O or N .

Ex. H_2O , H_2O_2 , H_2O_3 , H_2O_4 .

Intermolecular Forces

- (i) **London Dispersion Forces**
These are the forces of attraction between the molecules of all the substances. They are the weakest of the intermolecular forces.
- (ii) **Dipole-Dipole Forces**
These are the forces of attraction between the molecules of the polar substances. They are the strongest of the intermolecular forces.
- (iii) **Hydrogen Bonding**
This is the force of attraction between the molecules of the substances having electronegative atoms like F , O or N .

Intermolecular

forces are weakly and long-range, all these molecules are grouped as forces of the nature of different intermolecular forces are intermolecular. E.g., H_2 , O_2 , N_2 .

Hydrogen Bonding

It occurs when every hydrogen atom is bonded with a highly electronegative atom like F , O or N .

Intermolecular Forces

These are the forces of attraction between the molecules of all the substances. They are the weakest of the intermolecular forces.

Hydrogen Bonding

This is the force of attraction between the molecules of the substances having electronegative atoms like F , O or N .

Chemical Bonding Final Review Guide

Lei Huang

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