

CHEMICAL KINETICS

Reaction Rate

$$R = \frac{dA}{dt} \quad R_t = -\frac{d[A]}{dt} = \text{slope of conc. - time}$$



$$R_t = -\frac{1}{a} \frac{d[A]}{dt} = -\frac{1}{b} \frac{d[B]}{dt} = \frac{1}{c} \frac{d[C]}{dt}$$

Differential Rate Law

Reaction Rate \propto Conc. of React.

$$\text{Rate} \propto [A]^m [B]^n$$

$$R = k[A]^m [B]^n$$

$$\frac{d[A]}{a \ dt} = -k[A]^m [B]^n$$

Rate Constant

$$[A] = [B] = 1 \text{ M} \text{ then Rate} = k$$

Order

$$(m + n) = \text{overall order}$$

zero Order Rxn

Rate independent of conc. of react⁺

$$k = \frac{[A]_0 - [A]}{t} \text{ time}^{-1}$$

$$k = -\text{slope} \quad \text{y-axis: } [A]_0 - [A] \quad \text{x-axis: } t$$

$$t_{1/2} = \frac{[A]_0}{2k}$$

first Order Rxn

Rate depends linearly with conc.

$$k = \frac{1}{t} \ln \frac{[A]_0}{[A]} = \frac{2.303}{t} \log \frac{a_0}{a}$$

$$t_{1/2} = \frac{0.693}{k}$$

$$\ln[A] = -kt + \ln[A]_0$$

$$\ln k = -\text{slope} = \frac{1}{t} \ln \frac{[A]_0}{[A]}$$

$$\log[A] = \log[A]_0 - \frac{kt}{0.303}$$

$$k = \frac{-\text{slope}}{0.303} = \frac{0.303}{t} \log \frac{[A]_0}{[A]}$$

for plot w/o $\log \frac{[A]_0}{[A]}$ vs time

$$\text{slope} = \frac{1}{0.303}$$

Gas Phase 1st Order Rxn

$$k = 2.303 \log \frac{P_f - P_i}{t (P_f - P_i)}$$

$$= \frac{2.303}{t} \log \frac{P_f}{(2P_i - P_f)}$$

Temp. Dependence

$$E_a = E(\text{threshold}) - E(\text{reactant})_{\text{molecule}}$$

Arrhenius Eqⁿ

$$k = A e^{-E_a/RT}$$

- Exothermic: $\Delta E = -ve$

- Endothermic: $\Delta E = +ve$

$$\log k = \log A - \frac{E_a}{2.303R} \times \frac{1}{T}$$

$$\text{slope} = \frac{-E_a}{2.303R}$$

$$\log \frac{k_2}{k_1} = \frac{E_a}{2.303R} \left[\frac{1}{T_1} - \frac{1}{T_2} \right]$$

Chemical Kinetic Solution Manual

Raymond Chang

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[Chemistry](#) William Joseph Pietro,1990 [Physical Chemistry for the Biosciences](#) Raymond Chang,2005-02-11 This book is ideal for use in a one semester introductory course in physical chemistry for students of life sciences The author s aim is to emphasize the understanding of physical concepts rather than focus on precise mathematical development or on actual experimental details Subsequently only basic skills of differential and integral calculus are required for understanding the equations The end of chapter problems have both physiochemical and biological applications

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web documents to be submitted 1 the personal information provided in this form will be used solely for the purpose of transfer request 2 the school will not disclose the reason s for selection or non selection of student 3 this application is deemed to be unsuccessful if you do not hear from the school

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