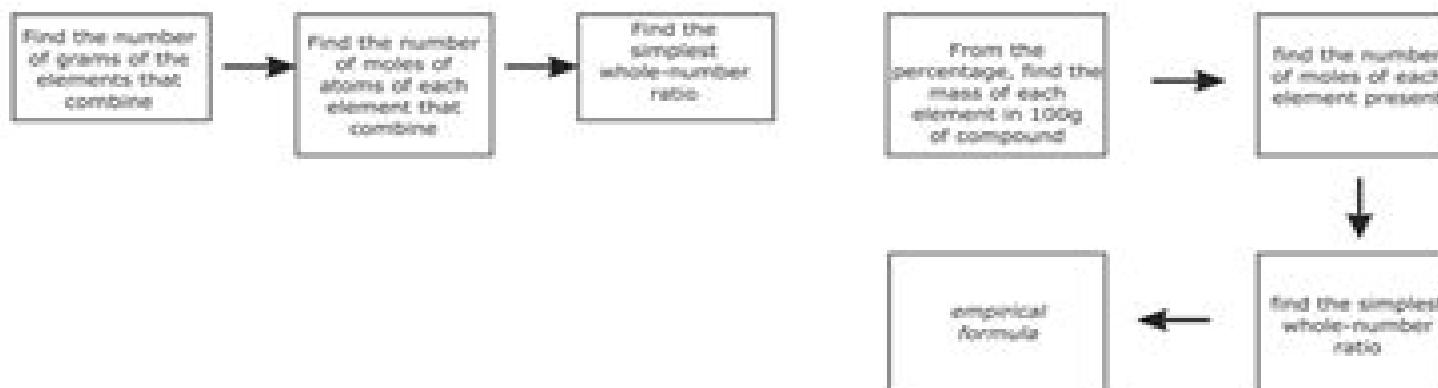


RATIOS ...	Carbon ($A_r = 12$)	Hydrogen ($A_r = 1$)	Chlorine ($A_r = 35.5$)	Comments and tips
Reacting mass or % mass	24.24	4.04	71.72	just think of it as based on 100g
molar ratio (mass in g / A_r)	$24.24 / 12 = 2.02$ mol	$4.04 / 1 = 4.04$ mol	$71.72 / 35.5 = 2.02$ mol	can now divide by smallest ratio number
simplest whole number ratio	$2.02 / 2.02 = 1$	$4.04 / 2.02 = 2$	$2.02 / 2.02 = 1$	
therefore the simplest ratio = empirical formula for the chlorinated hydrocarbon = CH_2Cl				
BUT the molecular mass is 99, and the empirical formula mass is 49.5 ($12+2+35.5$)				
AND $99 / 49.5 = 2$, and so the molecular formula must be $2 \times \text{CH}_2\text{Cl} = \text{C}_2\text{H}_4\text{Cl}_2$				



Chemistry Notes Form 3

Walter Gillian Clippinger

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Journal of Chemical Education, 1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers **Health & Drugs** Nicolae Sfetcu, 2014-05-02 Information about drugs side effects and abuse Drug prescription medication and therapy online stores to buy drugs Testing interaction administration and treatments for the health care Medicine is the branch of health science and the sector of public life concerned with maintaining or restoring human health through the study diagnosis treatment and possible prevention of disease and injury It is both an area of knowledge a science of body systems their diseases and treatment and the applied practice of that knowledge A drug is any biological substance synthetic or non synthetic that is taken for non dietary needs It is usually synthesized outside of an organism but introduced into an organism to produce its action That is when taken into the organisms body it will produce some effects or alter some bodily functions such as relieving symptoms curing diseases or used as preventive medicine or any other purposes School Science and Mathematics, 1926

Student Relationships Walter Gillian Clippinger, 1927 **Parliamentary Papers** Great Britain. Parliament. House of Commons, 1879 Certificate Chemistry Form 3, **Reports from Committees** Great Britain. Parliament. House of Commons, 1879 **Manual of the Public Instructions Acts and Regulations of the Council of Public Instruction of Nova Scotia** Nova Scotia, Nova Scotia. Council of Public Instruction, 1921 The Chemical News and Journal of Physical Science, 1913 Scientific American, 1903 **Research in Education**, 1972 "The" Athenaeum, 1845 Select Notes Mary Abby Thaxter Peloubet, 1892 **Journal of Education and School World**, 1889 The International Bookseller, 1892 **International Bookseller**, 1892 Chemical News and Journal of Industrial Science, 1872 **The American Catalogue**, 1880 Hoping to avoid the worry of caring for a valuable object Miss Teaberry gives away the locket her cat finds in the garden only to find herself enmeshed in a situation of escalating chaos **The Journal of Education**, 1899

Electrochemistry in Nonaqueous Solutions Kosuke Izutsu, 2002-05-06 Nonaqueous solutions are equally indispensable to electrochemistry Here Kosuke Izutsu brilliantly illustrates the numerous aspects of this fascinating topic whether the focus be on physicochemical processes or analytical methods The author discusses solvation and solvent effects emphasizing dynamic aspects important reactions including ionic and supercritical media as well as advanced techniques in polarography and voltammetry Throughout he effortlessly manages to provide a comprehensive overview while also presenting the very latest developments A number of example applications further enhance the practical value of this book and give it the feel of a reference work Written for both users and specialists this volume represents a wealth of vital information and belongs on every bookshelf

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