

Intro to Chemistry Final Exam Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- The Fahrenheit temperature scale is defined by which of the following temperatures?
 - Water boils at 100 degrees and freezes at 32 degrees.
 - Water boils at 212 degrees and freezes at 32 degrees.
 - Water evaporates at 212 degrees and freezes at 0 degrees.
 - Liquid water turns to a gas at 100 degrees and to a solid at 0 degrees.
- A cold-blooded reptile basks on a warm rock in the sun. Its body is warmed by
 - radiation.
 - conduction.
 - convection.
 - Both (a) and (b)
- Temperature is a measure of the average _____ energy of an object's particles.
 - mechanical
 - kinetic
 - potential
 - light
- The temperature at which all molecular motion stops is
 - 0°C.
 - 0°F.
 - 0 K.
 - All of the above
- Convert 500°F to degrees Celsius.
 - 260°C
 - 296°C
 - 84.2°C
 - 95.8°C
- Convert 300 K to the Celsius scale.
 - 0°C
 - 27°C
 - 73°C
 - 573°C
- Convert 458 K to the equivalent Fahrenheit temperature.
 - 185°F
 - 237°F
 - 365°F
 - 731°F
- Which of the following temperatures is impossible to measure?
 - 85°F
 - 50°C
 - 20 K
 - 545°F
- What is -173°C on the Kelvin scale?
 - 76 K
 - 89 K
 - 98 K
 - 448 K
- The principles of atomic theory recognized today were conceived by
 - Avogadro.
 - Bohr.
 - Dalton.
 - Rutherford.
- The total number of protons and neutrons in the nucleus of an atom is its
 - atomic number.
 - Avogadro number.
 - mass number.
 - average atomic mass.
- All atoms of the same element have the same
 - atomic mass.
 - number of neutrons.
 - mass number.
 - atomic number.
- Atoms of the same element can differ in
 - chemical properties.
 - mass number.
 - atomic number.
 - number of protons and electrons.
- Chlorine has atomic number 17 and mass number 35. It has
 - 17 protons, 17 electrons, and 18 neutrons.
 - 35 protons, 35 electrons, and 17 neutrons.
 - 17 protons, 17 electrons, and 52 neutrons.
 - 18 protons, 18 electrons, and 17 neutrons.
- Phosphorus-33 (atomic number 15) contains
 - 33 protons.
 - 18 neutrons.
 - 33 neutrons.
 - 18 protons.
- Neon-22 contains 12 neutrons. It also contains
 - 12 protons.
 - 22 protons.
 - 22 electrons.
 - 10 protons.
- Valence electrons determine an atom's

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