

# Chemistry Study Guide

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## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

## Matter:

Anything that has **mass** and **occupies space** ✖️🌐

Examples: air, gases, liquids, solids!

## Atoms:

Atoms are the **smallest, indivisible particle** 🧪👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	$p^+$	1	+1	Nucleus
Neutron	$n$	1	0	Nucleus
Electron	$e^-$	$\sim 0$	-1	Orbit/Shell

# Chemistry Hrw Study Guide

**Joacim Rocklöv**



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Student's Guide to Masterton and Slowinski's Chemical Principles Ray Boyington,William L. Masterton,1977      Science Education in Canadian Schools Graham W. F. Orpwood,Science Council of Canada,1984 Volume 1 of a three volume study of science education in Canadian schools describing the philosophy and methodology of the study and including an analysis of science textbooks used in Canadian schools The introduction presents criticisms and objectives of the study the scope and strategy the research conducted an analysis of science curriculum policies and the official aims and strategies      American Laboratory ,1970      **Photographer's Market** Melissa Milar,2005      **Holt Physical** Holt, Rinehart and Winston Staff,1994

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