

Chemistry Study Guide

Periodic Table of the Elements

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<small>19</small> K <small>39.10</small>	<small>20</small> Ca <small>40.08</small>	<small>21</small> Sc <small>44.96</small>	<small>22</small> Ti <small>47.88</small>	<small>23</small> V <small>50.94</small>	<small>24</small> Cr <small>52.00</small>	<small>25</small> Mn <small>54.94</small>	<small>26</small> Fe <small>55.85</small>	<small>27</small> Co <small>58.93</small>	<small>28</small> Ni <small>58.71</small>	<small>29</small> Cu <small>63.55</small>	<small>30</small> Zn <small>65.39</small>	<small>31</small> Ga <small>69.72</small>	<small>32</small> Ge <small>72.64</small>	<small>33</small> As <small>74.92</small>	<small>34</small> Se <small>78.96</small>	<small>35</small> Br <small>79.90</small>	<small>36</small> Kr <small>83.80</small>																														
<small>37</small> Rb <small>85.47</small>	<small>38</small> Sr <small>87.62</small>	<small>39</small> Y <small>88.91</small>	<small>40</small> Zr <small>91.22</small>	<small>41</small> Nb <small>92.91</small>	<small>42</small> Mo <small>95.94</small>	<small>43</small> Tc <small>98.91</small>	<small>44</small> Ru <small>101.07</small>	<small>45</small> Rh <small>102.91</small>	<small>46</small> Pd <small>106.42</small>	<small>47</small> Ag <small>107.87</small>	<small>48</small> Cd <small>112.41</small>	<small>49</small> In <small>114.82</small>	<small>50</small> Sn <small>118.71</small>	<small>51</small> Sb <small>121.76</small>	<small>52</small> Te <small>127.60</small>	<small>53</small> I <small>126.91</small>	<small>54</small> Xe <small>131.29</small>																														
<small>55</small> Cs <small>132.91</small>	<small>56</small> Ba <small>137.33</small>			<small>57</small> Hf <small>178.49</small>	<small>58</small> Ta <small>180.95</small>	<small>59</small> W <small>183.84</small>	<small>60</small> Re <small>186.21</small>	<small>61</small> Os <small>190.23</small>	<small>62</small> Ir <small>192.22</small>	<small>63</small> Pt <small>195.08</small>	<small>64</small> Au <small>196.97</small>	<small>65</small> Hg <small>200.59</small>	<small>66</small> Tl <small>204.38</small>	<small>67</small> Pb <small>207.2</small>	<small>68</small> Bi <small>208.98</small>	<small>69</small> Po <small>209</small>	<small>70</small> At <small>210</small>	<small>71</small> Rn <small>222</small>																													
<small>87</small> Fr <small>223</small>	<small>88</small> Ra <small>226</small>			<small>89</small> Rf <small>261</small>	<small>90</small> Db <small>262</small>	<small>91</small> Sg <small>266</small>	<small>92</small> Bh <small>264</small>	<small>93</small> Hs <small>277</small>	<small>94</small> Mt <small>268</small>	<small>95</small> Ds <small>271</small>	<small>96</small> Rg <small>272</small>	<small>97</small> Cn <small>285</small>	<small>98</small> Uut <small>289</small>	<small>99</small> Fl <small>288</small>	<small>100</small> Uup <small>289</small>	<small>101</small> Lv <small>293</small>	<small>102</small> Uus <small>289</small>	<small>103</small> Uuo <small>293</small>																													
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

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Table of Contents Chemistry Msl Study Guide

1. Understanding the eBook Chemistry Msl Study Guide
 - The Rise of Digital Reading Chemistry Msl Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Msl Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Msl Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Msl Study Guide
 - Personalized Recommendations
 - Chemistry Msl Study Guide User Reviews and Ratings
 - Chemistry Msl Study Guide and Bestseller Lists
5. Accessing Chemistry Msl Study Guide Free and Paid eBooks
 - Chemistry Msl Study Guide Public Domain eBooks
 - Chemistry Msl Study Guide eBook Subscription Services
 - Chemistry Msl Study Guide Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
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 - Chemistry Msl Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Msl Study Guide
 - Highlighting and Note-Taking Chemistry Msl Study Guide
 - Interactive Elements Chemistry Msl Study Guide
8. Staying Engaged with Chemistry Msl Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Msl Study Guide
9. Balancing eBooks and Physical Books Chemistry Msl Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Msl Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Msl Study Guide
 - Setting Reading Goals Chemistry Msl Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Msl Study Guide
 - Fact-Checking eBook Content of Chemistry Msl Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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