

Unit 6 Study Guide: Chemical Reactions

Friday 8th

- Know the signs that a chemical change has occurred
- Identify the different symbols in a chemical equation
- Know the difference between reactants and products
- Be able to write a chemical equation if given the word equation
- Be able to count atoms on both sides of a chemical equation
- Be able to balance chemical equations
- Identify the 5 types of reactions
- Be able to predict products for the 5 types of reactions
- Use solubility rules to predict states of matter for products in a double replacement reaction
- Be able to write complete and net ionic equations and identify spectator ions

In-class review

- Parts of a chemical equation, symbols, substances
- Signs of chemical change
- Balance
- Types of reaction
- Predicting products
 - Solubility rules (SNAP and Group 1)
 - Activity series for single replacement
 - Net ionic/complete ionic
 - Be able to write formulas if given word equations

1. Know the signs that a chemical change has occurred

- A precipitate is formed
- Gas is released (bubbles)
- Temperature change
- Odor/smell

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